

SUMMATIVE ASSESSMENT-II, 2014
SCIENCE
Class X

Time : 3½ hrs.

M.M. : 90

Date – 08.2.2014

General Instructions :

- The question paper comprises of two Sections, A and B. You are to attempt both the sections.
- All questions are compulsory.
- There is no choice in any of the questions.
- All questions of Section-A and all questions of Section-B are to be attempted separately.
- Question numbers 1 to 3 in Section-A are one mark questions. These are to be answered in one word or in one sentence.
- Question numbers 4 to 7 in Section-A are two marks questions. These are to be answered in about 30 words each.
- Question numbers 8 to 19 in Section-A are three marks questions. These are to be answered in about 50 words each.
- Question numbers 20 to 24 in Section-A are five marks questions. These are to be answered in about 70 words each.
- Question numbers 25 to 42 in Section-B are multiple choice questions based on practical skills. Each question is a one mark question. You are to select one most appropriate response out of the four provided to you.

SECTION - A

- Q.1** Predict the maximum number of valence electrons possible for atoms in first period of periodic table. (1)
- Q.2** In a beetle population, the number of green beetles is more than blue and red beetles. Give a reason behind this situation. (1)
- Q.3** Name the source of energy from which an ecosystem gets energy. (1)
- Q.4** Recycling is considered as a welcome practice to deal with the environmental problems. Justify this statement with two arguments. (2)
- Q.5** What is Chipko Movement ? Why should we conserve forest ? (2)
- Q.6** A converging lens has a focal length of 250 mm. Calculate its power and express it according to sign convention. (2)
- Q.7** For plants like banana and oranges that have lost their capacity to produce seeds a different ways are used for increasing their number. Explain the method used for such plants. (2)
- Q.8** Write structure of ethyl alcohol (ethanol) and identify its functional group. Give chemical equation of its reaction with
(i) sodium (ii) excess conc H_2SO_4 at 443 K (3)
- Q.9** The atomic number of Na and Mg is 11 and 12 respectively and they belong to the same period. (3)
- (a) Which one would have smaller atomic size ?
- (b) Which one would be more electropositive ?
- (c) To which group would each one belong ?

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M Tight



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Decoding **Cbse Class 10 Sa2 Science Question Paper 2014 Filetype**: Revealing the Captivating Potential of Verbal Expression

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