

Enzyme and Temperature

Background

For the experiment an enzyme called catalase, breaking down hydrogen peroxide, was used. All enzymes are affected by temperature.

Introduction

Enzymes are proteins that increase the rate of a chemical reaction by lowering its activation energy. They are called biological catalysts. Enzymes are made up of amino acids. The shape of an enzyme is very important. If the shape is changed, the enzyme will not work. Temperature can change the shape of an enzyme. This is why enzymes only work at certain temperatures.

Key Concepts

Enzymes are affected by temperature. At the right temperature, they work best.

Procedure

Step 1	Step 2	Step 3
1. Add 5 ml of hydrogen peroxide to a test tube.	2. Add 5 ml of catalase solution to the test tube.	3. Observe the reaction.

Observations

Time	Observation	Conclusion
0 min		Reaction is starting.
1 min		Reaction is increasing.
2 min		Reaction is at its peak.

At 0 min, there were no bubbles. At 1 min, there were a few bubbles. At 2 min, there were many bubbles. This shows that the reaction is increasing over time. The reaction is at its peak at 2 min. After 2 min, the reaction will start to decrease.

Results

Time	Volume of Gas Produced (ml)	Temperature (°C)
0 min	0.0	25.0
1 min	1.5	25.0
2 min	3.0	25.0
3 min	2.5	25.0
4 min	1.0	25.0
5 min	0.5	25.0

Conclusion

Time	Volume of Gas Produced (ml)	Temperature (°C)
0 min	0.0	25.0
1 min	1.5	25.0
2 min	3.0	25.0
3 min	2.5	25.0
4 min	1.0	25.0
5 min	0.5	25.0

The reaction is at its peak at 2 min. After 2 min, the reaction will start to decrease. This shows that the reaction is affected by temperature.

Graph



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RM Cervero



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