

Suggested solution

Hydration of the top alkene with aqueous acid goes via a carbocation intermediate, and would therefore give the product on the left. To get the product on the right we need a more sophisticated method, involving hydroboration to place hydroxyl on the less hindered carbon and then oxidation to the alcohol (p. 444 of the textbook).



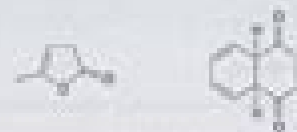
Using sodium borohydride to add two hydroxyl groups to the second alkene will give the diastereoisomer on the left, because the mechanism ensures both OH groups add to the same side of the alkene (p. 443 of the textbook). To get the trans diol, we can make an epoxide with *m*-CPBA and then open it with water, as full reaction then proceeds with inversion of configuration.



A possible alternative would be to start with the cis alkene the product will be the diol shown, which, if we culture it in an open chain conformation, is the same diol as the one that comes from the trans alkene via the epoxide.

**Suggested solutions for Chapter 20****20****PROBLEM 1**

Draw all possible enol forms of these carbonyl compounds, and comment on your stability.

**Purpose of the problem**

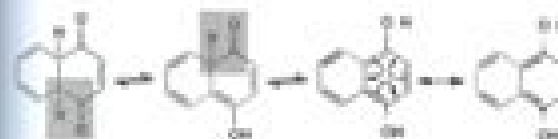
Simple exercise in drawing made with an extra twist.

Suggested solution

There is only one enol for the first compound and it might be stable because it is aromatic (two alkenes and one oxygen giving two electrons each, making six in all).



The second compound has seven possibilities, one of which is very stable indeed as it has two benzene rings. We haven't drawn the mechanism for such enolisation this time but note the different reaction arrows for tautomerisation (equilibrium) and dehydratation (two ways of drawing the same compound).



■ Aromaticity is discussed in chapter 17 of the textbook (pages 1021–103). Compounds in which lone pairs contribute to the aromatic system are introduced on page 102, but after that then whole chapters devoted to them (11 pages), 18 and 19. These chapters can be more advanced than you need at this stage.

Clayden Organic Chemistry Solutions Manual

Ian Pickup



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