

During this activity, you will use dialysis tubing and various solutions to study the movement of molecules across a selectively permeable membrane via diffusion and osmosis. Cellulose dialysis tubing is used as the selectively permeable membrane.

Pre-laboratory Questions

1. How are osmosis and diffusion similar? How are they different?
2. If a solution is hypertonic to a cell, in which direction will water move, and why?
3. If a solution is hypotonic to a cell, in which direction will water move, and why?

Materials

Station 1:	8 cups iodine-potassium iodide 70% glucose solution 2% starch solution 2 pairs of scissors	deionized water 6 plastic dropping pipets white thread 16 glucose test paper strips 2 rulers 2 permanent markers
Station 2:	dialysis tubing 2 pairs of scissors	2 rulers 250-mL beaker of deionized water
Station 3:	8 cups 20% glucose solution 10% glucose solution deionized water 2 pairs of scissors	6 plastic dropping pipets red thread blue thread paper towels 2 rulers 2 permanent markers
Station 4:	electronic balance(s) 16 plastic weigh boats	
Station 5:	dialysis tubing 2 pairs of scissors	2 rulers 250-mL beaker of deionized water

Carolina Study Guide Diffusion And Osmosis

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The Enigmatic Realm of **Carolina Study Guide Diffusion And Osmosis**: Unleashing the Language is Inner Magic

In a fast-paced digital era where connections and knowledge intertwine, the enigmatic realm of language reveals its inherent magic. Its capacity to stir emotions, ignite contemplation, and catalyze profound transformations is nothing lacking extraordinary. Within the captivating pages of **Carolina Study Guide Diffusion And Osmosis** a literary masterpiece penned by way of a renowned author, readers attempt a transformative journey, unlocking the secrets and untapped potential embedded within each word. In this evaluation, we shall explore the book's core themes, assess its distinct writing style, and delve into its lasting effect on the hearts and minds of people who partake in its reading experience.

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