

Computer Literacy Exam Information and Study Guide

The Computer Literacy Exam consists of two parts: an objective assessment and a hands-on assessment of Microsoft Office 2007 skills. The objective assessment consists of multiple choice questions. The objective segment tests the student's knowledge of Microsoft applications that include: Word, Excel, PowerPoint, and Access. The hands-on segment consists of four sections:

- **WORD:** Student will complete one or more documents utilizing Microsoft Word 2007.
- **EXCEL:** Student will complete a spreadsheet and chart utilizing Microsoft Excel 2007.
- **POWERPOINT:** Student will construct a slide show utilizing Microsoft PowerPoint 2007.
- **ACCESS:** Student will create a database, query, and report utilizing Microsoft Access 2007.

After completing both portions of the exam, the student will receive the results within **15 school days**.

The student is allowed four (4) hours for the total exam. If a student prefers, due to schedule conflicts, the test may be scheduled in two segments: a one (1) hour segment for the objective assessment and a three (3) hour segment for the hands-on assessment. **Testing arrangements must be made before the student begins the exam and may not be changed once the student has started the test.** In other words, a student may not sign up for a four-hour segment and then decide during the test time that he/she prefers to complete the hands-on segment of the test at a later time. Schedule an appointment by calling the Learning Assistance Center, Tutoring and Testing at 329-3876 or 3870.

If you are waiving the requirement, the fee for the test must be paid as follows: 1) when you call to make the appointment, a \$15 non-refundable, non-transferable administration fee to secure the appointment and 2) \$30 on the date of the test. If you are taking the test for credit, 1) pay a \$15 non-refundable, non-transferable administration fee to secure the appointment and 2) 40% of tuition for 3 credit hours prior to taking the test. Please arrive at the Learning Assistance Center a little early to allow time for processing paperwork.

In order to show eligibility to take the Computer Literacy exam, all UIW students must process a *Request for Credit by Exam* form through their advisor and the Assistant Registrar. Any students wanting credit hours must also process the *Application to Challenge* form. AD/CaP students should process the *Request for Credit by Examination* (to waive or for credit) and the *Application for Challenge Exam* (for credit) through their advisor and the Assistant Registrar for AD/CaP. On the day of the test, the student must bring the yellow copy of the approved *Request for Credit by Exam*, the *Application to Challenge* (if applicable), and the receipt of payment for the administration fee and the test. **The test will not be administered without these documents.**

NOTE: The test may only be taken once. Students who register for and complete the Competency Exam are not allowed to then challenge the Computer Literacy Course for credit. Students wishing to challenge any course for credit must first register with the Registrar's office and pay the appropriate fees for challenging a course.

This study guide includes the course content of a basic Computer Literacy course. Before registering for the test, students should review the study guide. The test will assess the student's knowledge of this material. A student must pass each segment of the test with a 70% or better to be considered to have satisfied the University of the Incarnate Word's requirement for computer literacy and be allowed to waive the requirement of a computer literacy course.

The following texts are recommended references for review purposes:

Microsoft Office 2007, Volume I, Volume 1, Third Edition, Grauer, et al. Prentice Hall, Upper Saddle River, NJ, Columbus, OH

Computer Literacy Exam Study Guide

Tobias Bleicker



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web $m \times c \times \Delta t$ equation 1 where m is the total mass of the solution solute plus solvent c is the specific heat of the solution

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web the goal of my experiment and research is to evaluate the various number of compounds to be used as hand warmers and locate the ones that can raise the temperature to 20 degrees celsius in 100 g water and costs as little as possible

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web in this lab you are challenged to use chemistry to determine which ionic compound makes the best hand warmer in this experiment you will collect data that will allow you to calculate the heat of solution ΔH_{soln} or the amount of energy released or absorbed in kJ/mol of the substance being dissolved

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web mar 4 2014 contains water the hand warmer is struck in a manner that ruptures the inner pouch releasing the ionic salt into the water of the outer pouch the salt dissolves and the water warms this type of hand warmer tends to produce a more vigorous heat than the dry powder type of hand warmer but does not produce heat for quite as long

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