CHAPTER 11 PRACTICE QUESTIONS KEY

INSTRUCTORS: Do not release this key until noon 4/22/18.

Use the following to answer questions 1-3:

Consider the representations below to answer the next three questions.



- How many of the following statements are correct concerning drawing 1?
 - Each molecule induces a dipole onto the next molecule in close proximity.
 - II. The phenomenon shown is relatively weak and short-lived.
 - III. C₈H₁₈ contains this type of interaction.
 - IV. The forces that exist in this example are London dispersion forces.
 - a) 0 b) 1 c) 2 d) 3 e) 4
- Which drawing best represents the interactions in a sample of HF?
 - a) I
 - b) II
 - d) IIId) I, II
 - e) all of the above
- 3. Which of the following statements are incorrect concerning drawing III?
 - a) Electrostatic interactions exist between the molecules.
 - The molecules find the best compromise between attraction and repulsion.
 - c) These molecules exhibit ionic bonding.
 - d) OCS exhibits this type of interaction
 - e) Two of the above statements are incorrect.
- 4. Which of the following is the correct order of boiling points for KNO₃, CH₃OH, C₃H₆, Ne?
 - a) Ne < CH₂OH < C₂H₆ < KNO₃
 - b) KNO₃ < CH₃OH < C₂H₆ < Ne</p>
 - e) Ne < C₂H₆ < KNO₃ < CH₃OH
 - d) Ne < C₂H₆ < CH₃OH < KNO₃
 - e) C₂H₆ < Ne < CH₂OH < KNO₃
- 5. What is the strongest intermolecular force (IMF) found in a sample of thousands of HBr molecules?
 - a London b. dipole dipole c. nonpolar d. H bond force e. polar

Chemistry 11 Practice Exam With Answers

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