

Acids, Bases & pH Review - Honors Chemistry

Directions: For all multiple choice questions, highlight the correct answer(s). Answer short answer questions in the box provided.

Part 1: Properties & definition of Acids & Bases

1. Complete the table below.

	ACIDS	BASES
Taste		
pH range		
Conducting electricity		
Neutralize		
Litmus		
Phenolphthalein		

2. An indicator

- is a type of acid.
- is a type of base.
- has a neutral pH.
- changes color depending on the pH.

3. Litmus and phenolphthalein are indicators. What does that mean?

4. Acids and bases "dissociate" in water to form H^+ (for acids) and OH^- (for bases). What does "dissociate" mean in this context?

5. Identify whether each substance is an acid or a base.

- HBr :
- HCO_2H :
- $Al(OH)_3$:
- $NH_3 + H_2O \rightarrow NH_4^+ + OH^-$; NH_3 is an _____ :
- $H_2SO_4 + H_2O \rightarrow H_3O^+ + HSO_4^-$; H_2SO_4 is an _____ :
- $HNO_3 \rightarrow H^+ + NO_3^-$; HNO_3 is an _____ :

Chemistry Study Guide Questions Acids And Bases

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Table of Contents Chemistry Study Guide Questions Acids And Bases

1. Understanding the eBook Chemistry Study Guide Questions Acids And Bases
 - The Rise of Digital Reading Chemistry Study Guide Questions Acids And Bases
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Study Guide Questions Acids And Bases
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry Study Guide Questions Acids And Bases
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Study Guide Questions Acids And Bases
 - Personalized Recommendations
 - Chemistry Study Guide Questions Acids And Bases User Reviews and Ratings
 - Chemistry Study Guide Questions Acids And Bases and Bestseller Lists
5. Accessing Chemistry Study Guide Questions Acids And Bases Free and Paid eBooks

- Chemistry Study Guide Questions Acids And Bases Public Domain eBooks
 - Chemistry Study Guide Questions Acids And Bases eBook Subscription Services
 - Chemistry Study Guide Questions Acids And Bases Budget-Friendly Options
6. Navigating Chemistry Study Guide Questions Acids And Bases eBook Formats
 - ePub, PDF, MOBI, and More
 - Chemistry Study Guide Questions Acids And Bases Compatibility with Devices
 - Chemistry Study Guide Questions Acids And Bases Enhanced eBook Features
 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemistry Study Guide Questions Acids And Bases
 - Highlighting and Note-Taking Chemistry Study Guide Questions Acids And Bases
 - Interactive Elements Chemistry Study Guide Questions Acids And Bases
 8. Staying Engaged with Chemistry Study Guide Questions Acids And Bases
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry Study Guide Questions Acids And Bases
 9. Balancing eBooks and Physical Books Chemistry Study Guide Questions Acids And Bases
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry Study Guide Questions Acids And Bases
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Chemistry Study Guide Questions Acids And Bases
 - Setting Reading Goals Chemistry Study Guide Questions Acids And Bases
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Chemistry Study Guide Questions Acids And Bases
 - Fact-Checking eBook Content of Chemistry Study Guide Questions Acids And Bases
 - Distinguishing Credible Sources
 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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