



CHEMISTRY

CHEAT SHEET



NAMING SIMPLE COMPOUNDS

NAMING SIMPLE COMPOUNDS CHEAT SHEET

Naming Compounds with Their Formulas
(Show the name corresponding to the formula)



Directions for naming:

1. Name the cation first, then the anion.
2. For ionic compounds, the cation name is the same as the element name.
3. For covalent compounds, the first element is named as is, and the second element is named with a prefix and an -ide suffix.
4. The order of the elements in the formula is the same as the order in the name.
5. The prefix for the number of atoms of each element is shown in parentheses.
6. The name of the compound is written with the elements in the same order as the formula.

Binary Compounds

First Element	Anion
H ₂	hydrogen
Li	lithium
Na	sodium
K	potassium
Rb	rubidium
Cs	cesium
Fr	francium
Be	beryllium
Mg	magnesium
Ca	calcium
Sc	scandium
Ti	titanium
V	vanadium
Cr	chromium
Mn	manganese
Fe	iron
Co	cobalt
Ni	nickel
Cu	copper
Zn	zinc
Ga	gallium
Ge	germanium
As	arsenic
Se	seelenium
Br	bromine
I	iodine
At	astatine

Common Ions (aq)

Formula	Name
H ⁺	hydrogen
Li ⁺	lithium
Na ⁺	sodium
K ⁺	potassium
Rb ⁺	rubidium
Cs ⁺	cesium
Fr ⁺	francium
Be ²⁺	beryllium
Mg ²⁺	magnesium
Ca ²⁺	calcium
Sc ³⁺	scandium
Ti ⁴⁺	titanium
V ⁵⁺	vanadium
Cr ⁶⁺	chromium
Mn ⁷⁺	manganese
Fe ³⁺	iron
Co ³⁺	cobalt
Ni ²⁺	nickel
Cu ²⁺	copper
Zn ²⁺	zinc
Ga ³⁺	gallium
Ge ⁴⁺	germanium
As ³⁺	arsenic
Se ⁴⁺	seelenium
Br ⁻	bromine
I ⁻	iodine
At ⁻	astatine

Naming Compounds with a Chemical and Formula



Carbon Dioxide

Directions for naming:

1. Name the cation first, then the anion.
2. For ionic compounds, the cation name is the same as the element name.
3. For covalent compounds, the first element is named as is, and the second element is named with a prefix and an -ide suffix.
4. The order of the elements in the formula is the same as the order in the name.
5. The prefix for the number of atoms of each element is shown in parentheses.
6. The name of the compound is written with the elements in the same order as the formula.

Binary Compounds

Element	Chemical Formula	Element	Chemical Formula
Hydrogen	H ₂	Carbon	C
Lithium	Li	Nitrogen	N
Sodium	Na	Oxygen	O
Potassium	K	Fluorine	F
Rubidium	Rb	Chlorine	Cl
Cesium	Cs	Bromine	Br
Francium	Fr	Iodine	I
Beryllium	Be	Astatine	At
Magnesium	Mg		
Calcium	Ca		
Scandium	Sc		
Titanium	Ti		
Vanadium	V		
Chromium	Cr		
Manganese	Mn		
Iron	Fe		
Cobalt	Co		
Nickel	Ni		
Copper	Cu		
Zinc	Zn		
Gallium	Ga		
Germanium	Ge		
Arsenic	As		
Seelenium	Se		
Bromine	Br		
Iodine	I		
Astatine	At		

Common Ions for Specific Elements

Element	Common Ions	Element	Common Ions
Hydrogen	H ⁺	Carbon	C ⁴⁺
Lithium	Li ⁺	Nitrogen	N ³⁻
Sodium	Na ⁺	Oxygen	O ²⁻
Potassium	K ⁺	Fluorine	F ⁻
Rubidium	Rb ⁺	Chlorine	Cl ⁻
Cesium	Cs ⁺	Bromine	Br ⁻
Francium	Fr ⁺	Iodine	I ⁻
Beryllium	Be ²⁺	Astatine	At ⁻
Magnesium	Mg ²⁺		
Calcium	Ca ²⁺		
Scandium	Sc ³⁺		
Titanium	Ti ⁴⁺		
Vanadium	V ⁵⁺		
Chromium	Cr ⁶⁺		
Manganese	Mn ⁷⁺		
Iron	Fe ³⁺		
Cobalt	Co ³⁺		
Nickel	Ni ²⁺		
Copper	Cu ²⁺		
Zinc	Zn ²⁺		
Gallium	Ga ³⁺		
Germanium	Ge ⁴⁺		
Arsenic	As ³⁺		
Seelenium	Se ⁴⁺		
Bromine	Br ⁻		
Iodine	I ⁻		
Astatine	At ⁻		

Cheat Sheet For Naming Compounds

CO Houle



Cheat Sheet For Naming Compounds:

Unveiling the Magic of Words: A Overview of "**Cheat Sheet For Naming Compounds**"

In some sort of defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their ability to kindle emotions, provoke contemplation, and ignite transformative change is actually awe-inspiring. Enter the realm of "**Cheat Sheet For Naming Compounds**," a mesmerizing literary masterpiece penned by a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve into the book is central themes, examine its distinctive writing style, and assess its profound effect on the souls of its readers.

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