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web you might need periodic table the first two ionization energies for beryllium are shown below be g be g e be g be 2 g e i  $1\,900\,$  kj mol i  $2\,1757\,$  kj mol which of the following identifies the most probable value for the third ionization energy for be and provides the best justification

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web the purpose of this lab is to introduce students to periodic trends it gives students an opportunity to explore the values of different properties for all elements in the periodic table to determine which properties of the elements follow a periodic trend

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web 1 how does atomic radius change as you move from left to right across a period explain your answer in terms of the number of protons in the nucleus 2 how does atomic radius change as you move from top to bottom in a group explain your answer in terms of the number of electron shells 3 on the representation of the periodic table below dry lab periodic trends answer key name studocu - Aug 07 2023

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