
CLEP Chemistry questions and answers with solutions 2024

John Dalton - ANSWER Proposed that matter is composed of atoms; these atoms have different identities called elements, which combine to form compounds; measured masses of reactants and products.

J.J. Thompson - ANSWER Observed deflection of particles in a cathode ray tube; proposed that atoms are composed of positive and negative charges; developed the plum pudding model of the atom.

Robert Millikan - ANSWER Calculated the charge-to-mass ratio of electrons using oil drops falling in an electric field; surmised the charge of a single electron.

Ernest Rutherford - ANSWER Used the deflection of alpha particles in a cathode ray tube to discover that most of the atom is empty space, with protons and neutrons centered in the nucleus.

Niels Bohr - ANSWER Determined that electrons exist around the nucleus at a fixed radius; electrons with higher energy exist farther from the nucleus. Electrons give off electromagnetic radiation when moving between energy levels.

Max Planck - ANSWER Determined that energy is quantized, or composed of discrete bundles.

$6.63 \times 10^{-34} \text{ J} \cdot \text{sec}$ - ANSWER Planck's Constant (h)

$3.00 \times 10^8 \text{ m/sec}$ - ANSWER Speed of Light (c)

$E = h\nu$ - ANSWER Energy of a Photon Formula (1)

$E = hc / \text{wavelength}$ - ANSWER Energy of a Photon Formula (2)

Louis DeBroglie - ANSWER Combined Einstein's relationship between mass and energy and the relationship between velocity and the wavelength of light. All particles with momentum have a corresponding wave nature.

Clep Chemistry Questions

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