INTERVIEWS QUESTIONS : CHEMISTRY

CHEMISTRY OBJECTIVE QUESTIONS

CHEMISTRY Multiple Choice Questions

1. The nucleus of an atom consists of

- A. electrons and neutrons
- B. electrons and protons
- C. protons and neutrons
- D. All of the above

2. The number of moles of solute present in 1 kg of a solvent is called its

- A. molality
- B. molarity
- C. normality
- D. formality

3. The most electronegative element among the following is

- A. sodium
- B. bromine
- C. fluorine
- D. oxygen

4. The metal used to recover copper from a solution of copper sulphate is

- A. Na
- B. Ag
- C. Hg
- D. Fe

5. The number of d-electrons in Fe2+ (Z = 26) is not equal to that of

- A. p-electrons in Ne(Z = 10)
- B. s-electrons in Mg(Z = 12)
- C. d-electrons in Fe(Z = 26)
- D. p-electrons in CI(Z = 17)

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Prof. S. K. Khanna, Dr. N. K. Verma, Dr. B. Kapila

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