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Acetic Acid (pKa = 4.75) titrated with NaOH

a. 0.100 M / volume 10.00 mL / pH 3.75 / 0.100 M

b. 0.100 M / volume 10.00 mL / pH 4.75 / 0.100 M

c. 0.100 M / pH 4.75 / compound that releases acetate
0.100 M / 10.00 mL / 0.100 M / 0.100 M

Acetic Acid, H^+ Acetate, H^+ Acetate, and H^+ Acetate

Acetic Acid / Acetate / Acetic Acid / Acetate
Acetic Acid / Acetate / Acetic Acid / Acetate

Species	Add (M, Base (M, or both (M)	Acetate with (M) (M or both)	Conductivity (C or both)
H^+	H^+	H^+	H^+
H^+	H^+	H^+	H^+
H^+	H^+	H^+	H^+
H^+	H^+	H^+	H^+

Chemistry Note Taking Guide Episode 1101 Answers

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