

Introduction to the Combined Gas Law

- The Combined Gas Law is a combination of Boyle's Law, Charles' Law, and Gay-Lussac's Law.

- It utilizes three factors – _____ and _____

P_1 = Initial Pressure

V_1 = Initial Volume

T_1 = Initial Temperature

$$\frac{P_1 V_1}{T_1} = \frac{P_2 V_2}{T_2}$$

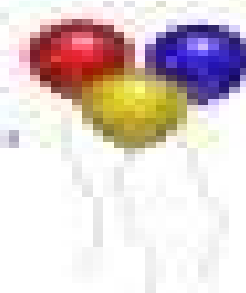
P_2 = Final Pressure

V_2 = Final Volume

T_2 = Final Temperature

Example Questions

- 1) The gas in a balloon has a pressure of 2.4 atm and a temperature of 22.0 K. The balloon is taken outside for HNO₃, which causes the pressure to increase to 4.8 atm. If the initial volume was 1.5 L, what was final volume of the gas?



- 2) A deflated football contains 0.5 L of gas at a pressure of 1000 mmHg. The football is then filled with gas until it contains 1.5 L of gas at 7000 mmHg of pressure. If the gas, released at a temperature of 22.0 K, what was the final temperature?



Combined Gas Law Study Guide Answersy

Karen Timberlake



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Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual Leo J. Malone, Theodore O. Dolter, 2012-01-03 The 9th edition of Malone s Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections assessment exercises at the end each section and relevant chapter problems at the end of each chapter A new Math Check allows quick access to the needed basic skill The first chapter now includes brief introductions to several fundamental chemical concepts and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem Every concept in the text is clearly illustrated with one or more step by step examples Making it Real essays have been updated to present timely and engaging real world applications emphasizing the relevance of the material they are learning This edition continues the end of chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter

Chemical Principles Study Guide/Solutions Manual John Krenos, Joseph Potenza, Carl Hoeger, 2007-01-18 Written for general chemistry courses Chemical Principles helps students develop chemical insight by showing the connection between chemical principles and their applications

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to promote active learning through a variety of exercises with answers and mastery exams Also contains complete solutions to odd numbered problems Chemical Principles Student's Study Guide & Solutions Manual John Krenos,Joseph Potenza,Loretta Jones,Lynn Koplitz,Thomas Spence,2004-03-19 This combination manual is designed to help students avoid common mistakes and understand the material better The solutions manual section includes detailed answers and explanations to the odd numbered exercises in the text **The Complete Chemistry Study Guide and Note Cards and MCAT** Konstantinos Papadopoulos,2012-07-06 *Student Study Guide and Solutions Manual to Accompany General, Organic, and Biochemistry* Katherine J. Denniston,Denniston Katherine,2006 *Barron's Science 360: A Complete Study Guide to Chemistry with Online Practice* Barron's Educational Series,Mark Kernion,Joseph A. Mascetta,2021-09-07 provides a complete guide to the fundamentals of chemistry Page 4 of cover

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