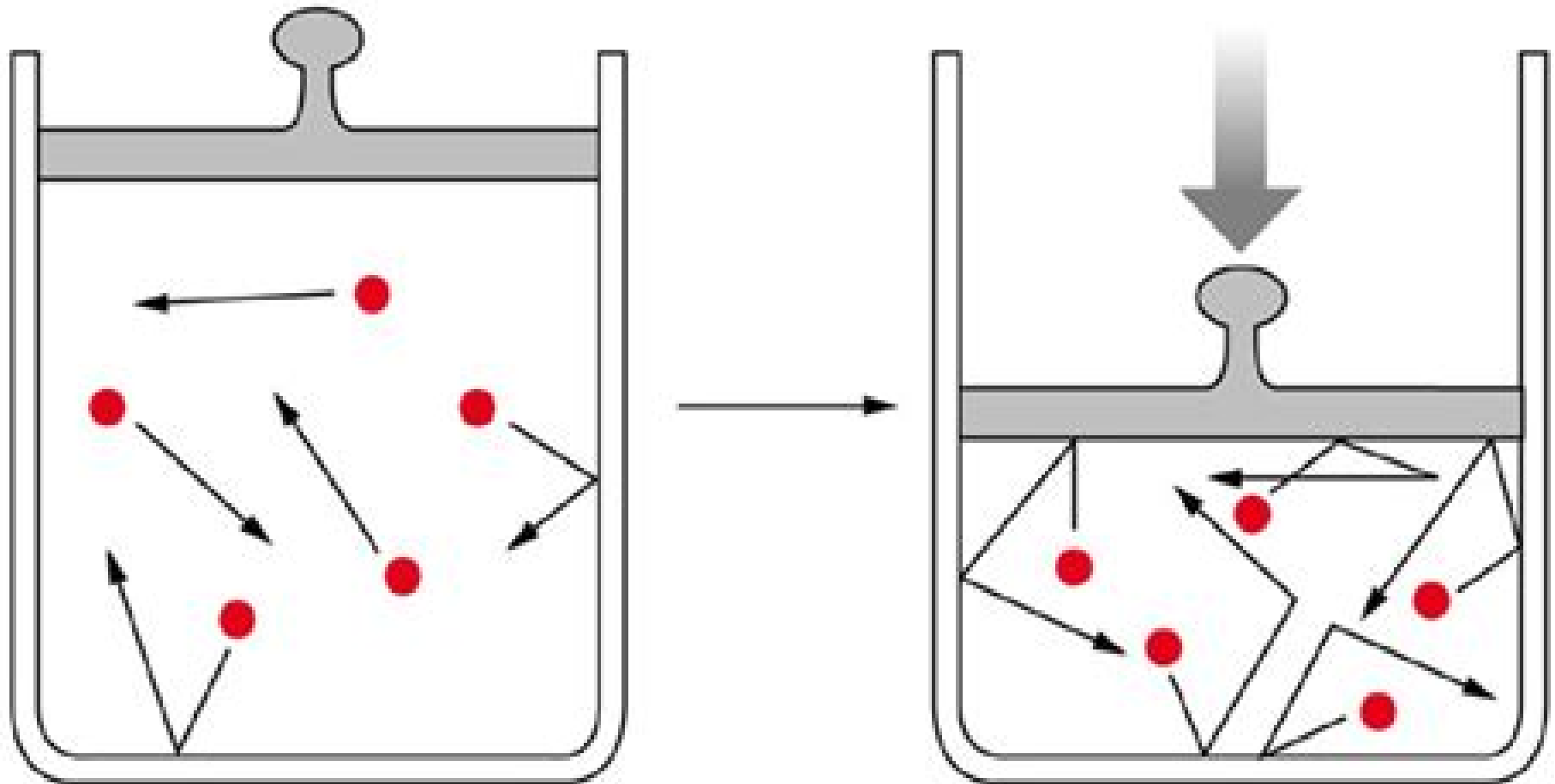


Boyle's Law: $P_1V_1 = P_2V_2$

Decreasing volume increases collisions and increases pressure.



$V_1 = 1.0 \text{ L}$
 $P_1 = 100 \text{ mm Hg}$

$V_2 = 0.5 \text{ L}$
 $P_2 = 200 \text{ mm Hg}$

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**Lee R. Summerlin, James Lee
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Library of Congress Catalog: Motion Pictures and Filmstrips Library of Congress, 1968 **Chemical Demonstrations** Lee R. Summerlin, James Lee Ealy, Christie L. Borgford, Julie B. Ealy, 1988 Chemical demonstrations L R Summerlin v 2 **NBS Special Publication**, 1974 *The Colliery Engineer and Metal Miner*, 1895 **Chemistry Decoded** Leonard W. Fine, 1976 **Cd+Update+ Contemp College Physics** Jones, 2002 *General Organic and Biological Chemistry* Kenneth W. Raymond, 2009-12-14 This general organic and biochemistry text has been written for students preparing for careers in health related fields such as nursing dental hygiene nutrition medical technology and occupational therapy It is also suited for students majoring in other fields where it is important to have an understanding of the basics of chemistry Students need have no previous background in chemistry but should possess basic math skills The text features numerous helpful problems and learning features **Introductory Chemistry** Otto William Nitz, 1961

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