

*Determination
and Use of*
**STABILITY
CONSTANTS**

SECOND EDITION

Arthur E. Martell
Ramunas J. Motekaitis

Determination And Use Of Stability Constants

Xiang Xie



Determination And Use Of Stability Constants:

Determination and Use of Stability Constants Arthur Earl Martell, Ramunas J. Motekaitis, 1992 This book describes potentiometric methods for determining stability constants and explains how these constants can be used to describe metal ion speciation in complex environmental and biological systems It also provides three original computer programs on a disk for calculating stability constants and for using stability constants to calculate concentrations of molecular species in solution The author gives examples of calculations for simple metal chelates for metal complexes of large organic molecules and for mixtures containing several metal ions and complexing agents in aqueous solution They also describe common errors in calculating stability constants and how to avoid them This carefully revised second edition is now even more useful to the reader and in particular to those who make use of the program disk Each program has been revised to improve speed control and error trapping

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Martell, Ramunas J. Motekaitis, 1988 **Physical Inorganic Chemistry** S. F. A. Kettle, 2013-11-11 GEORGE CHRISTOU Indiana University Bloomington I am no doubt representative of a large number of current inorganic chemists in having obtained my undergraduate and postgraduate degrees in the 1970s It was during this period that I began my continuing love affair with this subject and the fact that it happened while I was a student in an organic laboratory is beside the point I was always enchanted by the more physical aspects of inorganic chemistry while being captivated from an early stage by the synthetic side and the measure of creation with a small c that it entails I nevertheless found the application of various theoretical spectroscopic and physicochemical techniques to inorganic compounds to be fascinating stimulating educational and downright exciting The various bonding theories for example and their use to explain or interpret spectroscopic observations were more or less universally accepted as belonging within the realm of inorganic chemistry and textbooks of the day had whole sections on bonding theories magnetism kinetics electron transfer mechanisms and so on However things

changed and subsequent inorganic chemistry teaching texts tended to emphasize the more synthetic and descriptive side of the field. There are a number of reasons for this and they no doubt include the rise of diamagnetic organometallic chemistry as the dominant subdiscipline within inorganic chemistry and its relative narrowness vis à vis physical methods required for its prosecution.

Metal Complexes in Aqueous Solutions Arthur E. Martell, Robert D. Hancock, 2013-06-29

Stability constants are fundamental to understanding the behavior of metal ions in aqueous solution. Such understanding is important in a wide variety of areas such as metal ions in biology, biomedical applications, metal ions in the environment, extraction, metallurgy, food chemistry, and metal ions in many industrial processes. In spite of this importance, it appears that many inorganic chemists have lost an appreciation for the importance of stability constants and the thermodynamic aspects of complex formation with attention focused over the last thirty years on newer areas such as organometallic chemistry. This book is an attempt to show the richness of chemistry that can be revealed by stability constants when measured as part of an overall strategy aimed at understanding the complexing properties of a particular ligand or metal ion. Thus, for example, there are numerous crystal structures of the Li ion with crown ethers. What do these indicate to us about the chemistry of Li with crown ethers? In fact, most of these crystal structures are in a sense misleading in that the Li ion forms no complexes or at best very weak complexes with familiar crown ethers such as 12-crown-4 in any known solvent. Thus, without the stability constants, our understanding of the chemistry of a metal ion with any particular ligand must be regarded as incomplete. In this book, we attempt to show how stability constants can reveal factors in ligand design which could not readily be deduced from any other physical technique.

Stability Constants, 1988

Nuclear Science Abstracts, 1970

NSA is a comprehensive collection of international nuclear science and technology literature for the period 1948 through 1976, pre-dating the prestigious INIS database which began in 1970. NSA existed as a printed product. Volumes 1-33 initially created by DOE's predecessor, the U.S. Atomic Energy Commission (AEC). NSA includes citations to scientific and technical reports from the AEC, the U.S. Energy Research and Development Administration and its contractors, plus other agencies and international organizations, universities, and industrial and research organizations. References to books, conference proceedings, papers, patents, dissertations, engineering drawings, and journal articles from worldwide sources are also included. Abstracts and full text are provided if available.

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