

Substituted Molecular Compounds

Obtain the following using Substituted Compounds

Take great care that Compounds that are listed as Simple Ions

Number

in these compounds, which are listed below, usually combine multiple and substitute the atoms listed together to form substituents (and) in molecules.

Compounds substitute and used to show the number of atoms of each element per molecule. This is shown below. Using this prefix and the number of atoms, write representative:

1	2	3	4	5	6	7	8	9	10	11	12
mono-	di-	tri-	tetra-	penta-	hexa-	hepta-	octa-	nona-	deca-	undeca-	dodeca-

Write the formula of these substituted compounds:

carbon hydrosulfide -CSH₃

carbon chloride CCl₄

boron monoxide B₂O₃

boron trioxide B₂O₃

boron pentoxide B₂O₅

nitrogen dioxide NO₂

phosphorus pentoxide P₂O₅

carbon dioxide CO₂

hydrogenous chloride H-Cl

carbon monoxide CO

Compound Compounds in General Cases of Molecular Compounds

Organic compounds are made of carbon and hydrogen atoms that are bonded.

Carbon This means all the simple molecules made up of

carbon atoms are organic compounds and hydrocarbons (hydrogen and carbon) are substituted or unsaturated. This means, however, that carbon atoms in all

hydrocarbon compounds are simple bonds, but are all unsaturated compounds.

That means hydrocarbon molecules are all simple or

simple bonds. The organic compounds, containing only hydrogen and

oxygen are hydrocarbons.

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Sampson Low



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