Chemistry 338 2nd Semester Final Exam Review Sheet

Unit 1 - Matter & Measurement - Chapters 1 & 2

 A student experimentally determines the density of an object to be 4.2 g/mL. If the accepted value is 5.0 g/mL, calculate the % error.

% error = $\frac{experimentat-accepted}{accepted} \times 100$ % error = $\frac{4.2-5.0}{5.0} \times 100 = -16\%$ error

Unit 2 - Periodic Table - Chapter 5

2.

Element	Symbol	Atomic #	Protons	Neutrons	Electrons	Mass#	Charge
Fungsten ion	194 W+2	7-4	74	110	72	184	+2

 Using the following atomic radius values, predict the atomic radius of potassium (K): Li = 145 pm, Na = 180 pm, K = 7, Rb = 235 pm K = 220 pm

Unit 3 - Reactions/Compounds - Chapters 7 & 8

- 4. Classify the compound as ionic or covalent. Name the compound
 - Fe₂O₃ iron (III) oxide (ionic roman numerals for transition metals, change ending to -ide)
 - b. Cl₂O₈ dichlorine octaoxide (covalent use prefixes, change ending to -ide)
- 5. Classify the compound as ionic or covalent. Write the formula
 - a. Trisulfur Dinitride S₃N₂ (covalent)
 - b. Zinc Hydroxide Zn(OH)2 (ionic look up ion charges and criss-cross) Zn⁻² OH⁻³
 - c. Chromium (III) nitrate Cr(NO₃)₃ (ionic criss-cross) Cr²³ NO₃⁻¹
- 6. Write the equation, balance it, and classify the type of reaction:

Iron (II) sulfate reacts with sodium hydroxide to yield iron(II) hydroxide and sodium sulfate

> 1 FeSO₄ + 2 NaOH → 1 Fe(OH)₂ + 1 Na₂SO₄ double displacement (two ionic compounds reacting)

7. Balance the equation and classify the type of reaction:

1 Cu + 2 AgNO₃ → 1 Cu(NO₃)₂ + 2 Ag Single displacement (element and a compound reacting)

Unit 4 - Moles - Chapter 3

8. Calculate the molar mass of Fe₂(SO₄)₅.

Fe (2)(55.9 g/mol) = 111.8 g/mol

S (3)(32.1 g/mol) = 96.3 g/mol

+ O (12)(16.0 g/mol) = +192 g/mol

= 400.1 g/mol

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Judith P. Foard

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fights off the pain by focusing on school and the girls basketball team When Ruth meets Maurice Langlois a French speaking Canadian immigrant he helps her forget about Rex As their love grows however they realize they will soon be separated by hundreds of miles when they leave for college Their attempt to resolve problems related to their separation gets even more complicated when Rex makes a surprising reappearance in Ruths life Will Ruth and Maurices love for each other survive

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