

Q3) Attempt any FOUR of the following : **[16]**

- a) Define the term cryoscopic constant. Derive thermodynamically the relation.

$$K_f = \frac{RT_0^2}{1000L_f}$$

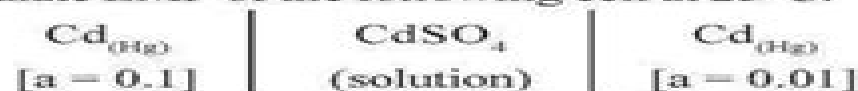
- b) Explain the effect of hydrogen bonding on the following physical properties.
- solubility and
 - Viscosity.
- c) Derive an expression for concentration cells with transference.
- d) A second order reaction where $a = b$, is 30% completed with in 500 seconds. How long will it take for the reaction to go to 70% completion?
- e) The osmotic pressure of a 5% by volume solution of a substance at 18°C is 6.64 atm. Calculate the molecular weight of the substance.
- f) Describe eutectic phase diagram for $\text{Zn} - \text{Cd}$ system.

Q4) Attempt any FOUR of the following : **[16]**

- a) What do you understand by 'geometrical isomerism'. Explain the use of E and Z nomenclature to describe geometrical isomers. Explain the stability of geometrical isomers.
- b) What is plane polarised light? How it is related to optical activity? Explain the use of polarimeter to measure optical activity.
- c) Explain the following types of organic reactions :
- Substitution reactions and
 - Oxidation and reduction reactions.
- d) Balance the following equation by oxidation number method :



- e) Calculate EMF of the following cell at 25°C .



- f) Findout the pH of a 0.002 M acetic acid solution if it is 2.3% ionised at this dilution.

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1. Understanding the eBook Chemistry 1 Fybsc University Question Paper 2014
 - The Rise of Digital Reading Chemistry 1 Fybsc University Question Paper 2014
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry 1 Fybsc University Question Paper 2014
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry 1 Fybsc University Question Paper 2014
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry 1 Fybsc University Question Paper 2014
 - Personalized Recommendations
 - Chemistry 1 Fybsc University Question Paper 2014 User Reviews and Ratings
 - Chemistry 1 Fybsc University Question Paper 2014 and Bestseller Lists
5. Accessing Chemistry 1 Fybsc University Question Paper 2014 Free and Paid eBooks
 - Chemistry 1 Fybsc University Question Paper 2014 Public Domain eBooks
 - Chemistry 1 Fybsc University Question Paper 2014 eBook Subscription Services
 - Chemistry 1 Fybsc University Question Paper 2014 Budget-Friendly Options

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 - ePub, PDF, MOBI, and More
 - Chemistry 1 Fybsc University Question Paper 2014 Compatibility with Devices
 - Chemistry 1 Fybsc University Question Paper 2014 Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemistry 1 Fybsc University Question Paper 2014
 - Highlighting and Note-Taking Chemistry 1 Fybsc University Question Paper 2014
 - Interactive Elements Chemistry 1 Fybsc University Question Paper 2014
8. Staying Engaged with Chemistry 1 Fybsc University Question Paper 2014
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry 1 Fybsc University Question Paper 2014
9. Balancing eBooks and Physical Books Chemistry 1 Fybsc University Question Paper 2014
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry 1 Fybsc University Question Paper 2014
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Chemistry 1 Fybsc University Question Paper 2014
 - Setting Reading Goals Chemistry 1 Fybsc University Question Paper 2014
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chemistry 1 Fybsc University Question Paper 2014
 - Fact-Checking eBook Content of Chemistry 1 Fybsc University Question Paper 2014
 - Distinguishing Credible Sources
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 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
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