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Chemistry 1411 Practice Exam 1, Chapters 1-3

Show your calculations clearly on show work problems. A periodic table is also included. Good luck!

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1	China there	mauron been of	significant	fluorumes un t	the follow	periodice management	being (25 total)	0

1) 6.022 X 10²³ atoms/mol 2) 0.049001 g

a) 3, 5

b) 4, 2

c) 23, 7

d) 4, 5

A rectangular block with a length, width, and height of 2.55 cm, 2.55 cm, and 1.44 cm has a density of 0.850 g/cm³. What is the mass of this object in grams? (3 pts)
 a) 5.46 g
 b) 7.96 g
 c) 8.65 g
 d) 10.4 g

(3 pts) Isotopes are atoms that:

a) have different atomic numbers

b) have the same mass number

c) have the same number of protons but different numbers of neutrons

d) have different numbers of electrons

4. A general summary based on observed and measurable behavior in many different systems

is called a: (3 pts)
a) hypothesis

b) natural law

c) model.

d) theory

What is the number of protons, neutrons, and electrons in the iron-56 atom? (3 pts)

a) 56 protons; 26 neutrons; 24 electrons b) 26 protons; 30 neutrons; 24 electrons

c) 24 protons; 56 neutrons; 26 electrons d) 26 protons, 30 neutrons; 26 electrons

Which of the following pairs of elements is a <u>transition element</u> and a <u>halogen</u>? (3 pts)

a) W and Si

b) Cr and Ar

c) Au and Br

d) Ba and U

Which of the following compounds is molecular? (3 pts)

a) BarCla

b) NaOH

c) SFa

d) KNOs

Which of the following elements is a noble gas? (3 pts)

a) Mg

b) Xe

e) Pb

d) Ber

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Joseph William Mellor

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Table of Contents Chemistry 1411 Lab Answers

- 1. Understanding the eBook Chemistry 1411 Lab Answers
 - The Rise of Digital Reading Chemistry 1411 Lab Answers
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Chemistry 1411 Lab Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry 1411 Lab Answers
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Chemistry 1411 Lab Answers
 - Personalized Recommendations
 - Chemistry 1411 Lab Answers User Reviews and Ratings
 - Chemistry 1411 Lab Answers and Bestseller Lists
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 - Highlighting and Note-Taking Chemistry 1411 Lab Answers
 - Interactive Elements Chemistry 1411 Lab Answers
- 8. Staying Engaged with Chemistry 1411 Lab Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry 1411 Lab Answers
- 9. Balancing eBooks and Physical Books Chemistry 1411 Lab Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry 1411 Lab Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Chemistry 1411 Lab Answers
 - Setting Reading Goals Chemistry 1411 Lab Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chemistry 1411 Lab Answers
 - Fact-Checking eBook Content of Chemistry 1411 Lab Answers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements

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