Study Guide-Chemistry Semester 1

Chapter 1

1. What are the 5 types of Chemistry? (page 8)

Organic Chemistry, Inorganic Chemistry, Biochemistry, Analytic Chemistry, Physical Chemistry.

- What is the difference between pure and applied chemistry? (page 9)
- 3. Name 2 areas in which chemistry improved your world? (page 12-17)
- 4. How did Lavoisier transform chemistry? (page 21)
- What are the Steps of the Scientific Method? (page 22)
- 6. What is a hypothesis? (page 22)
- 7. What is the difference between a scientific theory and a scientific law? (page 23)
- What are 3 steps in solving numeric problems? (page 29)
- What is the difference between a Manipulated Variable and a Responding Variable (page 22).

Chapter 2

- 1. What is the difference between a Heterogeneous Mixture and a Homogeneous Mixture?
- 2. What is the difference between a Chemical change and a chemical property?
- What is the Law of Conservation of Mass?

Chapter 3 - Scientific Measurement

- 1. State the SI units for length, mass, temperature, and time
- express numbers using scientific notation (for example, 3.29 x 10°)
- metric units conversions
 - a. be able to convert from meters (m) to km, cm, mm, μm, nm (p.74)
 - i. convert 0.044 km to meters
 - b. be able to convert from liters (L) to ml, cm³, and μl (p.75)
 - i. convert 15 cm3 in liters
 - be able to convert from kilograms (kg) to g, mg, and µg (p.76)
 - i. convert 750 dg in grams

Chapter 4 – Atomic Structure

- 1. Define atom.
- 2. State the 4 ideas of Dalton's atomic theory (p.102)
- 3. List the three types of subatomic particles (p.104)
- 4. What is the electrical charge of each subatomic particle?
- 5. Describe the structure of the atom according to the Rutherford model.
- 6. What distinguishes atoms of one element from atoms of a different element?
- What is the <u>atomic number</u>? What is the <u>mass number</u>?
- 8. How do you calculate the number of neutrons in an atom (p.111)?

80 Br

For the element neutrons? what is the atomic number, the mass number, and the number of

- 10. What is an isotope?
- 11. Why do isotopes have different mass numbers?
- 12. What is the correct name for a row across the periodic table?
- 13. What is the correct name for a vertical group down the periodic table?

Chemistry Study Guide Answer

T. W. Graham Solomons, Craig B. Fryhle, Scott A. Snyder

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