

UNIT 1: MATTER & MEASUREMENT

Experimental Design

Independent Variable = Manipulated condition
Dependent Variable = Responding variable
Controlled Variable = Must be same for all groups

As part of the trial, substances are mixed in 4 beakers at the same temperature, volume and pH, resulting in the following concentrations:



Independent Variable = HCl Concentration (molarity)
Dependent Variable = Reaction time
Controlled Variable = Temperature, Mg sample size

Physical vs. Chemical

Physical:

- * Substance retains its properties
- * Chemical formulas do not change
- * Usually Phase Change

Examples:



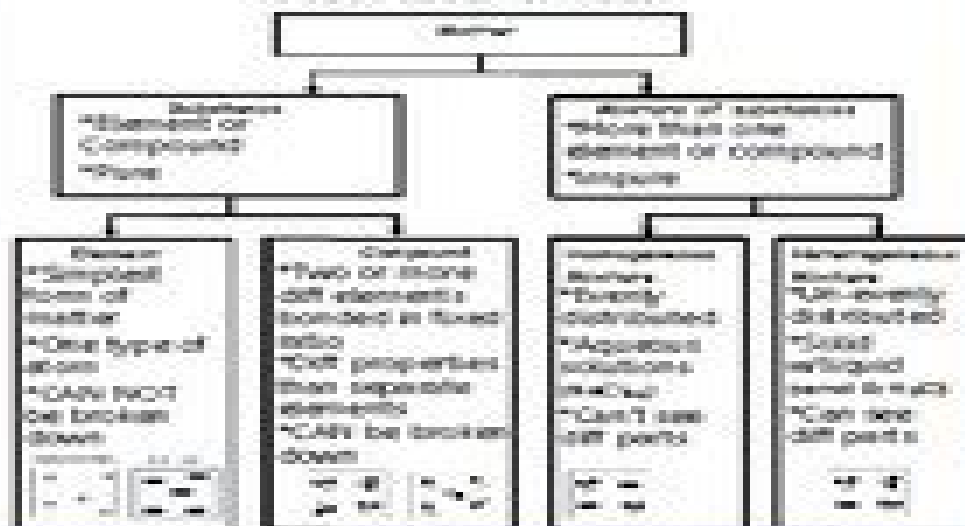
Chemical:

- * Properties do not change
- * Chemical formulas change
- * rust, burn, oxidize, combust, corrode

Examples:



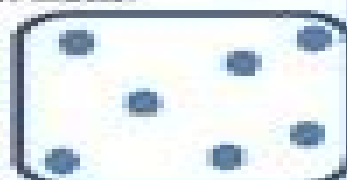
Classification of Matter



Phases of Matter

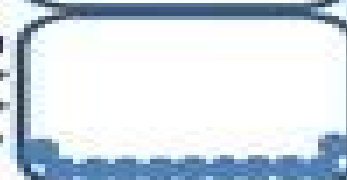
Gas

- NO definite volume
- NO definite shape
- Completely fills container
- Weak forces of attraction



Liquid

- definite volume
- NO definite shape
- flows and molds



Solid

- definite shape + volume
- regular geometric pattern
- vibrates in a fixed location



Physical Separation of Mixtures

Filtration

solid separated from liquid/solvents



key property:
solubility (solid vs liquid/solvent)

Distillation

liquids with different PB heated to boil

key property:
Boiling Point

Extraction

key property:
intermolecular forces

Chromatography

different color ink dye separated by solubility

key property:
Molecular Polarity

Significant Figures

Atlantic Pacific Rule

Decimal point is present or significant



Multiply/Divide

Answer must have least sig figs in data

$$\begin{array}{r} 24.68 \\ \times 3.250 \\ \hline 79.83 \\ 79.8 \end{array}$$

Chemistry Regents Survival Guide

SA Dillow



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Table of Contents Chemistry Regents Survival Guide

1. Understanding the eBook Chemistry Regents Survival Guide
 - The Rise of Digital Reading Chemistry Regents Survival Guide
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Regents Survival Guide
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry Regents Survival Guide
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Regents Survival Guide
 - Personalized Recommendations
 - Chemistry Regents Survival Guide User Reviews and Ratings
 - Chemistry Regents Survival Guide and Bestseller Lists
5. Accessing Chemistry Regents Survival Guide Free and Paid eBooks
 - Chemistry Regents Survival Guide Public Domain eBooks
 - Chemistry Regents Survival Guide eBook Subscription Services
 - Chemistry Regents Survival Guide Budget-Friendly Options

6. Navigating Chemistry Regents Survival Guide eBook Formats
 - ePub, PDF, MOBI, and More
 - Chemistry Regents Survival Guide Compatibility with Devices
 - Chemistry Regents Survival Guide Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemistry Regents Survival Guide
 - Highlighting and Note-Taking Chemistry Regents Survival Guide
 - Interactive Elements Chemistry Regents Survival Guide
8. Staying Engaged with Chemistry Regents Survival Guide
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry Regents Survival Guide
9. Balancing eBooks and Physical Books Chemistry Regents Survival Guide
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry Regents Survival Guide
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Chemistry Regents Survival Guide
 - Setting Reading Goals Chemistry Regents Survival Guide
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chemistry Regents Survival Guide
 - Fact-Checking eBook Content of Chemistry Regents Survival Guide
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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