MULTIPLE CHOICE QUESTIONS

When potassium dichro	mate crystals are heated wi	th hydrochloric acid, we get
(A) Oxygen	(B) Hydrogen	(C) Sulphur dioxide
(D) Hydrogen sulphide	(E) Chlorine	Ans : E
Which of the following	elements has lowest melting	point?
(A) Copper	(B) Niobium	(C) Molybdenum
(D) Tungsten	(E) Mercury	Ans : E
In acidic medium if post	assium permanganate is use	ed as an oxidising agent, we get
(A) MnO ₂ ²⁻ ,	(B) Mn ²	(C) MnO ₂
(D) Mn_2O_1	(E) None of these	Ans: B
Among the following oute oxidation state?	ermost configurations of trans	itition metals, which shows the highest
(A) $3d^34s^2$	(B) 3d ⁵ 4s ¹	(C) $3d^{5}4s^{2}$
(D) 3d ⁶ 4s ²	(E) 3d ⁷ 4s ²	Ans : C
Highest (+7) oxidation s	tate is shown by	
(A) Co	(B) Cr	(C) V
(D) Mn	(E) Ni	Ans : D
Potassium permanganate	on heating gives	
	$O_{s} = (B) MnO_{2s}, O_{s} KOH$	(C) K_2MnO_4, O_2

Chemistry Multiple Choice Questions And Answers

Arshad Iqbal

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