

Balancing and Reactions Test Unit 2 SCH3U

Name: _____

 K T C A **Assessment of Learning (Product)**
15 20 5 4

Multiple Choice: Circle the most correct answer. (10K)

1. Propane gas, C_3H_8 , burns in excess oxygen gas. When the equation for this reaction is correctly balanced and all coefficients are reduced to their lowest whole-number terms, the coefficient for H_2O is...

- A) 1 B) 2 C) 3 D) 4 E) 5 f) 6

2. State the type of reaction: $2 Na + 2 H_2O \rightarrow 2 NaOH + H_2$

- A) synthesis B) decomposition C) single displacement
D) double displacement E) combustion F) no reaction

3. Which of the following types of chemical reactions produces water and an ionic compound?

- A) single displacement B) neutralization C) decomposition D) synthesis

4. State the type of reaction: $Ca(NO_3)_2 + 2 HCl \rightarrow CaCl_2 + 2 HNO_3$

- A) synthesis B) decomposition C) single displacement
D) double displacement E) combustion F) no reaction

5. If chlorine gas is produced by single displacement, the other halogen in the reaction must be

- A) bromine B) iodine C) astatine D) fluorine

6. In what kind of reaction do two or more substances combine to form a new compound?

- A) decomposition reaction C) double-displacement reaction
B) single displacement D) synthesis reaction

7. A group whose atoms (M) undergo the reaction:



- A) group 1 B) group 2 C) group 13 D) group 14 E) group 15

8. When heated, a metal carbonate decomposes into a metal oxide and

- A) carbon. B) carbon dioxide C) oxygen. D) hydrogen.

9. Oxides of active metals, such as CaO , react with water to produce

- A) metal carbonates. B) metal hydrides. C) acids. D) metal hydroxides.

10. Which of the following statements is true about single-displacement reactions?

- A) They are restricted to metals. C) Two reactants produce two products.
B) They involve a single product. D) Any metal replaces any other metal.

Chemical Equations And Reactions Test Answers

L Cohen



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Table of Contents Chemical Equations And Reactions Test Answers

1. Understanding the eBook Chemical Equations And Reactions Test Answers
 - The Rise of Digital Reading Chemical Equations And Reactions Test Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemical Equations And Reactions Test Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemical Equations And Reactions Test Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemical Equations And Reactions Test Answers
 - Personalized Recommendations
 - Chemical Equations And Reactions Test Answers User Reviews and Ratings
 - Chemical Equations And Reactions Test Answers and Bestseller Lists
5. Accessing Chemical Equations And Reactions Test Answers Free and Paid eBooks
 - Chemical Equations And Reactions Test Answers Public Domain eBooks
 - Chemical Equations And Reactions Test Answers eBook Subscription Services
 - Chemical Equations And Reactions Test Answers Budget-Friendly Options

6. Navigating Chemical Equations And Reactions Test Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Chemical Equations And Reactions Test Answers Compatibility with Devices
 - Chemical Equations And Reactions Test Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemical Equations And Reactions Test Answers
 - Highlighting and Note-Taking Chemical Equations And Reactions Test Answers
 - Interactive Elements Chemical Equations And Reactions Test Answers
8. Staying Engaged with Chemical Equations And Reactions Test Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemical Equations And Reactions Test Answers
9. Balancing eBooks and Physical Books Chemical Equations And Reactions Test Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemical Equations And Reactions Test Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Chemical Equations And Reactions Test Answers
 - Setting Reading Goals Chemical Equations And Reactions Test Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chemical Equations And Reactions Test Answers
 - Fact-Checking eBook Content of Chemical Equations And Reactions Test Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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