

Name: _____

Date: _____

Student Exploration: Chemical Equations

Vocabulary: Avogadro's number, chemical equation, chemical formula, chemical reaction, coefficient, combination, combustion, conservation of matter, decomposition, double replacement, molar mass, mole, molecule, product, reactant, single replacement, subscript

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

1. A candle is placed on one pan of a balance, and an equal weight is placed on the other pan.

What would happen if you lit up the candle and waited for a while? _____

2. Suppose the candle was placed in a large, sealed jar that allowed it to burn for several minutes before running out of oxygen. The candle and jar are balanced by an equal weight.

In this situation, what would happen if you lit up the candle and waited? _____

Gizmo Warm-up

Burning is an example of a **chemical reaction**. The law of **conservation of matter** states that no atoms are created or destroyed in a chemical reaction. Therefore, a balanced **chemical equation** will show the same number of each type of atom on each side of the equation.

To set up an equation in the *Chemical Equations Gizmo™*, type the **chemical formulas** into the text boxes of the Gizmo. First, type in "H₂+O₂" in the **Reactants** box and "H₂O" in the **Products** box. This represents the reaction of hydrogen and oxygen gas to form water.

Reactants		Products
H ₂ +O ₂	→	H ₂ O

1. Check that the **Visual** display is chosen on each side of the Gizmo, and count the atoms.
 - A. How many hydrogen atoms are on the **Reactants** side? _____ **Products** side? _____
 - B. How many oxygen atoms are on the **Reactants** side? _____ **Products** side? _____
2. Based on what you see, is this equation currently balanced? _____

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Table of Contents Chemical Equation Explore Learning Gizmo Teacher Guide

1. Understanding the eBook Chemical Equation Explore Learning Gizmo Teacher Guide
 - The Rise of Digital Reading Chemical Equation Explore Learning Gizmo Teacher Guide
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemical Equation Explore Learning Gizmo Teacher Guide
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemical Equation Explore Learning Gizmo Teacher Guide
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemical Equation Explore Learning Gizmo Teacher Guide
 - Personalized Recommendations
 - Chemical Equation Explore Learning Gizmo Teacher Guide User Reviews and Ratings

- Chemical Equation Explore Learning Gizmo Teacher Guide and Bestseller Lists
- 5. Accessing Chemical Equation Explore Learning Gizmo Teacher Guide Free and Paid eBooks
 - Chemical Equation Explore Learning Gizmo Teacher Guide Public Domain eBooks
 - Chemical Equation Explore Learning Gizmo Teacher Guide eBook Subscription Services
 - Chemical Equation Explore Learning Gizmo Teacher Guide Budget-Friendly Options
- 6. Navigating Chemical Equation Explore Learning Gizmo Teacher Guide eBook Formats
 - ePub, PDF, MOBI, and More
 - Chemical Equation Explore Learning Gizmo Teacher Guide Compatibility with Devices
 - Chemical Equation Explore Learning Gizmo Teacher Guide Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemical Equation Explore Learning Gizmo Teacher Guide
 - Highlighting and Note-Taking Chemical Equation Explore Learning Gizmo Teacher Guide
 - Interactive Elements Chemical Equation Explore Learning Gizmo Teacher Guide
- 8. Staying Engaged with Chemical Equation Explore Learning Gizmo Teacher Guide
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemical Equation Explore Learning Gizmo Teacher Guide
- 9. Balancing eBooks and Physical Books Chemical Equation Explore Learning Gizmo Teacher Guide
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemical Equation Explore Learning Gizmo Teacher Guide
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Chemical Equation Explore Learning Gizmo Teacher Guide
 - Setting Reading Goals Chemical Equation Explore Learning Gizmo Teacher Guide
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chemical Equation Explore Learning Gizmo Teacher Guide
 - Fact-Checking eBook Content of Chemical Equation Explore Learning Gizmo Teacher Guide
 - Distinguishing Credible Sources

-
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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pédagogiques 21 5 2 former les enseignants et les soutenir 23 5 3 activités pilotes 25

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