

Chemistry 1212
Study Guide for Final Exam

1) Chapter 11 Liquids, Solids, and Intermolecular forces

a) Review Intermolecular Forces

$$1) E = \frac{1}{4\pi\epsilon_0} \frac{q_1 q_2}{r}$$

b) Dispersion forces, Dipole-Dipole, Hydrogen Bonding, Ion-Dipole, Surface Tension, Viscosity, Capillary Action.

c) Vaporization and Vapor Pressure, Dynamic Equilibrium

1) Clausius-Clapeyron Equations

$$a) \ln P_{\text{vap}} = \frac{-\Delta H_{\text{vap}}}{R} \left(\frac{1}{T} \right) + \ln \beta$$

$$b) \ln \frac{P_2}{P_1} = \frac{-\Delta H_{\text{vap}}}{R} \left(\frac{1}{T_2} - \frac{1}{T_1} \right) \quad \text{Two-point form}$$

d) Critical Point

e) Fusion and Sublimation

f) Phase Diagrams

g) Intermolecular Forces and Water

h) Band Theory (Crystalline Structures)

I recommend that you review your quizzes and try a couple of problems under each section of the Exercises.

2) Chapter 12 Solutions

a) Solutions

1) Types

2) Solubility of substances

3) Entropy as it relates to solubility

4) Effects of intermolecular forces on solubility

b) Energetics of solution formation

$$a) \Delta H_{\text{soln}} = \underbrace{\Delta H_{\text{solute}}}_{\text{endothermic (+)}} + \underbrace{\Delta H_{\text{solvent}}}_{\text{endothermic (+)}} + \underbrace{\Delta H_{\text{mix}}}_{\text{exothermic (-)}}$$

1) Heats of hydration (See the relationships on pages 446 & 447)

2) Solution Equilibrium

3) Solution Concentration Methods

4) Gas as a solute (Henry's Law)

5 Colligative Properties

a) Freezing point depression and Boiling point elevation

I recommend that you review your quizzes and try a couple of problems under each section of the Exercises.

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Table of Contents Chem 1212 Acs Final Exam Study Guide

1. Understanding the eBook Chem 1212 Acs Final Exam Study Guide
 - The Rise of Digital Reading Chem 1212 Acs Final Exam Study Guide
 - Advantages of eBooks Over Traditional Books
2. Identifying Chem 1212 Acs Final Exam Study Guide
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chem 1212 Acs Final Exam Study Guide
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chem 1212 Acs Final Exam Study Guide
 - Personalized Recommendations
 - Chem 1212 Acs Final Exam Study Guide User Reviews and Ratings
 - Chem 1212 Acs Final Exam Study Guide and Bestseller Lists

5. Accessing Chem 1212 Acs Final Exam Study Guide Free and Paid eBooks
 - Chem 1212 Acs Final Exam Study Guide Public Domain eBooks
 - Chem 1212 Acs Final Exam Study Guide eBook Subscription Services
 - Chem 1212 Acs Final Exam Study Guide Budget-Friendly Options
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 - Chem 1212 Acs Final Exam Study Guide Compatibility with Devices
 - Chem 1212 Acs Final Exam Study Guide Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chem 1212 Acs Final Exam Study Guide
 - Highlighting and Note-Taking Chem 1212 Acs Final Exam Study Guide
 - Interactive Elements Chem 1212 Acs Final Exam Study Guide
8. Staying Engaged with Chem 1212 Acs Final Exam Study Guide
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chem 1212 Acs Final Exam Study Guide
9. Balancing eBooks and Physical Books Chem 1212 Acs Final Exam Study Guide
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chem 1212 Acs Final Exam Study Guide
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Chem 1212 Acs Final Exam Study Guide
 - Setting Reading Goals Chem 1212 Acs Final Exam Study Guide
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chem 1212 Acs Final Exam Study Guide
 - Fact-Checking eBook Content of Chem 1212 Acs Final Exam Study Guide
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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