

I promise that I will abide by the Virginia Tech Honor Code while taking this test.

Signature _____

Printed Name _____

$c = 3.00 \times 10^8 \text{ m/s}$ $h = 6.63 \times 10^{-34} \text{ J}\cdot\text{s}$ Avogadro's Number = 6.02×10^{23}

The ionization energy for hydrogen, if the electron is in the $n = 1$ main energy level, is $2.18 \times 10^{-18} \text{ J}$.

$1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^6 5s^2 4d^{10} 5p^6 6s^2 4f^{14} 5d^{10} 6p^6 7s^2 5f^{14} 6d^{10}$

1. An aqueous solution is prepared by dissolving 20.0 grams of potassium sulfate, K_2SO_4 (FW = 174) in sufficient water to prepare 500 milliliters of solution. What is $[\text{K}^+]$, the molarity of K^+ ion, in this solution?

1. 0.115 M
2. 0.230 M
3. 0.460 M
4. 0.080 M

$$[\text{K}_2\text{SO}_4] = \frac{(20.0)}{(174)} = 0.230 \text{ M}$$



$$[\text{K}^+] = 2(0.230) = \underline{\underline{0.460 \text{ M}}}$$

2. At 25.0°C , the pOH of a solution is 4.75, what is $[\text{H}^+]$ in the solution?

1. $[\text{H}^+] = 9.25 \text{ M}$
2. $[\text{H}^+] = 1.78 \times 10^{-5} \text{ M}$
3. $[\text{H}^+] = 5.62 \times 10^{-10} \text{ M}$
4. $[\text{H}^+] = 9.61 \times 10^{-5} \text{ M}$

$$\text{pH} = 14 - 4.75 = 9.25$$

$$[\text{H}^+] = 10^{-9.25} = \underline{\underline{5.62 \times 10^{-10} \text{ M}}}$$

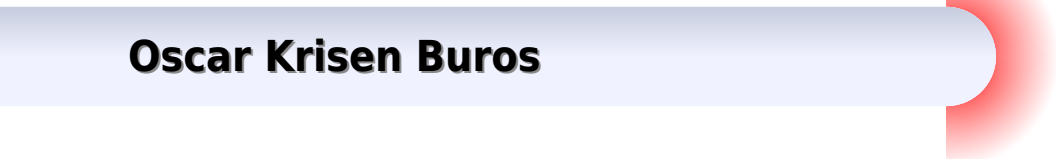
3. A compound of carbon and hydrogen is combusted completely with O_2 to produce CO_2 and H_2O . In a particular experiment, the combustion of a sample of the compound produces 5.50 grams of CO_2 and 4.50 grams of water. What is the mass of hydrogen in the sample that was combusted?

1. 1.50 grams H
2. 0.500 gram H
3. 0.250 gram H
4. 1.00 gram H

$$\frac{4.50 \text{ g H}_2\text{O}}{18.0 \text{ g H}_2\text{O}} \times \frac{1 \text{ mol H}_2\text{O}}{1 \text{ mol H}_2\text{O}} \times \frac{2 \text{ mol H}}{1 \text{ mol H}_2\text{O}} \times \frac{1.0 \text{ g H}}{1 \text{ mol H}} = \underline{\underline{0.500 \text{ g H}}}$$

Chem 1035 Placement Test

Oscar Krisen Buros



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Tests Richard C. Sweetland,1991 Revises the information in the second edition and presents over 700 new or revised tests The Psychology section contains 20 subsections Education has 54 subsections and Business has 13 subsections Does not contain reliability validity and normative data Use the complementary Test Critiques series for this information The ... Mental Measurements Yearbook Oscar Krisen Buros,1972 **Reinforcement for Modern Concrete Structures** Khan Mahmud Amanat,2025-03-26 There are a lot of excellent textbooks on the design of reinforced concrete structures However based on professional experience of more than thirty years the author feels that a separate book dedicated only to discuss the properties behavior and use of reinforcement in concrete construction is highly justified Conventional textbooks on concrete structures focus primarily on the mechanics of reinforced concrete design Properties of reinforcement are discussed in a limited manner only those which are deemed just adequate for discussing the theory of concrete mechanics Typically such books contain little or no background information or explanation as to why the various code provisions or rules regarding the reinforcement are imposed or formulated It is observed that the application of modern materials and technologies in reinforcement manufacturing does not get proper attention or consideration in the engineering practices of many professionals Based on the interaction with many engineers engaged in the design and construction of concrete structures the author feels that there is a lack of the essential background knowledge of the engineering properties and behavior of concrete reinforcement among many of them And such a lack of knowledge often leads to improper implementation of the code design provisions Limited scope of our existing engineering curricula is primarily responsible for this This book has been written with an aim to fill this gap and to make engineering students and practicing engineers more up to date **The Seventh Mental Measurements Yearbook** Price Stern Sloan Publishing,Buros Institute,Buros Center,1972 Customers who place a standing order for the Tests in Print series or the Mental Measurements Yearbook series will receive a 10% discount on every volume To place your standing order please call 800 755 1105 in the U S or 402 472 3581 outside the U S The most widely acclaimed reference series in education and psychology the Mental Measurements Yearbooks are designed to assist professionals in selecting and using standardized tests The series initiated in 1938 provides factual information critical reviews and comprehensive bibliographic references on the construction use and validity of all tests published in English The objectives of the Mental Measurements Yearbooks have remained essentially the same since the publication of the series These objectives include provision to test users of factual information on all known tests published as separates in the English speaking countries of the world candidly critical test reviews written for the MMYs by qualified professional people representing a variety of viewpoints excerpts of the critical portions of test reviews which have been published in professional journals comprehensive bibliographies for specific tests of references which have been examined for their relevance to the particular tests listing of books on measurements and closely related fields as well as excerpts of evaluative

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