

Chapter 10 Study Guide- Chemical Quantities

TEST: Thursday 1/27/11

10.1 The Mole: Measurement of Matter

Define the following terms:

- mole
- Avogadro's number
- representative particle
- molar mass

1. What kind of representative particles can one calculate using Avogadro's number?

Converting Moles to # of Particles $\leftarrow \rightarrow$ Converting # of Particles to Moles

$$1 \text{ Mole} = 6.02 \times 10^{23} \text{ particles}$$

therefore

or

$$\frac{1 \text{ mol}}{6.02 \times 10^{23} \text{ particles}}$$



(particles $\rightarrow$ moles)

*If you are given
particles and want to
find the moles you
divide by 6.02×10^{23}
particles.*

$$\frac{6.02 \times 10^{23} \text{ particles}}{1 \text{ mol}}$$



(moles $\rightarrow$ particles)

*If you are given
moles and you want
to find particles then
you multiply by 6.02×10^{23}
particles.*

Chemistry Section 10 Study Guide Chemical Quantities

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