

Quiz & Worksheet - Chemical Kinetics & Equilibrium

1. Which of the following statements is true about chemical equilibria in general?

- ☐ At EQ, the rate of the forward reaction is equal to the rate of the reverse reaction.
- ☐ Equilibrium is the result of the termination of all chemical activity.
- ☐ At EQ, the total concentration of products equals the total concentration of reactants, **products = reactants**.
- ☐ There is only one set of concentrations that equals K_c .

2. The equilibrium expression for the equilibrium constant shows

- ☐ the ratio of concentrations of products to concentrations of reactants at equilibrium.
- ☐ the temperature of an equilibrium system.
- ☐ the rates of the forward and reverse reactions.
- ☐ the time it takes to reach equilibrium.

3. If at equilibrium for a reversible reaction, the forward rate constant is 5.7×10^{-4} and the reverse rate constant is 4.5×10^{-3} , what is the K_c at for this equilibrium state?

- ☐ 0.13
- ☐ 7.2
- ☐ 1
- ☐ 1.8×10^2

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Chemistry Kinetics And Equilibrium Questions

James N. Jensen



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