#### Gases Test SCH3U

Name:

\_\_\_K \_\_T \_\_C \_\_A Assessment of Learning (Product)
15 9 5 10

#### Multiple Choice. Circle the most correct answer. (15K)

1. Convert the pressure 1.80 kPa to atm.

- A) 2 atm B) 115 atm C) 132 at
  - C) 132 atm D) 245 atm
- E) None of the above

2. What happens to the temperature of a gas when it is expanded quickly?



3. If all of the following flasks are the same size, at the same temperature, and contain the same number of molecules, in which flask will the pressure be highest?

- A) All have the same pressure
- B) Flask I

C) Flask 2

D) Flask 3

E) Flask 4

4. Which diagram best represents a gas in a closed container?









According to the kinetic molecular theory, the Kelvin temperature of an ideal gas is proportional to which one of the following:

A) acceleration

- B) momentum
- C) kinetic energy
- D) potential energy

6. A gas is \_\_\_\_\_ and assumes \_\_\_\_ of its container whereas a liquid is \_\_\_\_ and assumes \_\_\_\_ of its container.

- A) compressible, the volume and shape, not compressible, the shape of a portion
- B) compressible, the shape, not compressible, the volume and shape
- C) compressible, the volume and shape, compressible, the volume
- D) condensed, the volume and shape, condensed, the volume and shape
- E) condensed, the shape, compressible, the volume and shape

# **Chemistry Gases Unit Study Guide Answers**

**Jill Meryl Levy** 

#### **Chemistry Gases Unit Study Guide Answers:**

Organic Chemistry, Student Study Guide and Solutions Manual David R. Klein, 2017-01-04 This is the Student Study Guide and Solutions Manual to accompany Organic Chemistry 3e Organic Chemistry 3rd Edition is not merely a compilation of principles but rather it is a disciplined method of thought and analysis Success in organic chemistry requires mastery in two core aspects fundamental concepts and the skills needed to apply those concepts and solve problems Readers must learn to become proficient at approaching new situations methodically based on a repertoire of skills These skills are vital for successful problem solving in organic chemistry Existing textbooks provide extensive coverage of the principles but there is far less emphasis on the skills needed to actually solve problems Chemistry, Student Study Guide James E. Brady, Fred Senese, 2008-01-28 The image on the front cover depicts a carbon nanotube emerging from a glowing plasma of hydrogen and carbon as it forms around particles of a metal catalyst Carbon nanotubes are a recently discovered allotrope of carbon Three other allotropes of carbon buckyballs graphite and diamond are illustrated at the left as is the molecule methane CH4 from which nanotubes and buckyballs can be made The element carbon forms an amazing number of compounds with structures that follow from simple methane found in natural gas to the complex macromolecules that serve as the basis of life on our planet The study of chemistry also follows from the simple to the more complex and the strength of this text is that it enables students with varied backgrounds to proceed together to significant levels of achievement

Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual Leo J. Malone, Theodore O. Dolter, 2012-01-03 The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections assessment exercises at the end each section and relevant chapter problems at the end of each chapter A new Math Check allows guick access to the needed basic skill The first chapter now includes brief introductions to several fundamental chemical concepts and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem Every concept in the text is clearly illustrated with one or more step by step examples Making it Real essays have been updated to present timely and engaging real world applications emphasizing the relevance of the material they are learning This edition continues the end of chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter GED Test Prep Chemistry Review--Exambusters Flash Cards--Workbook 3 of 13 GED Exambusters, 2016-06-01 GED Prep Flashcard Workbook 3 CHEMISTRY 700 questions Essential chemistry formulas and concepts Topics Metric System Matter Atoms Formulas Moles Reactions Elements Chemical Bonds Phase Changes Solutions Reaction Rates Acids and Bases Oxidation and Reduction Introduction to Organic ADDITIONAL WORKBOOKS GED Prep Flashcard Workbook 9 COLLEGE PREP VOCABULARY Intermediate Level 2 350

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