

Date Monday 10/18/2021 Name Aya Turner
Section _____ Team _____
Instructor _____

Pre-Lab Study Questions

1. What is meant by the simplest formula of a compound?

The lowest whole number ratio of the atoms in the formula.

2. How does a hydrate differ from an anhydrate?

Hydrate is an ionic compound with water and anhydrate is an ionic compound without water.

3. What happens when a hydrate is heated?

When a hydrate is heated, it has the water removed and becomes an anhydrate.

4. A hydrate of CoCl_2 with a mass of 6.00 g is heated strongly. After cooling, the mass of the anhydrate is 3.27 g.

- a. How many grams of H_2O were lost from the hydrate?

$$\begin{array}{r} 6.00 \\ - 3.27 \\ \hline 2.73 \end{array} \quad 2.73 \text{ g } \text{H}_2\text{O}$$

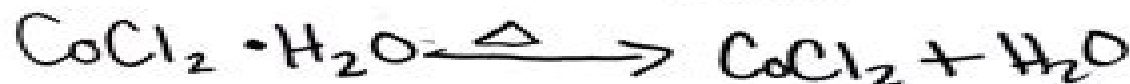
- b. What is the % water in the hydrate?

$$\frac{2.73}{6.00} \times 100 = 45.5\% \text{ of } \text{H}_2\text{O}$$

- c. What is the formula of the CoCl_2 hydrate?

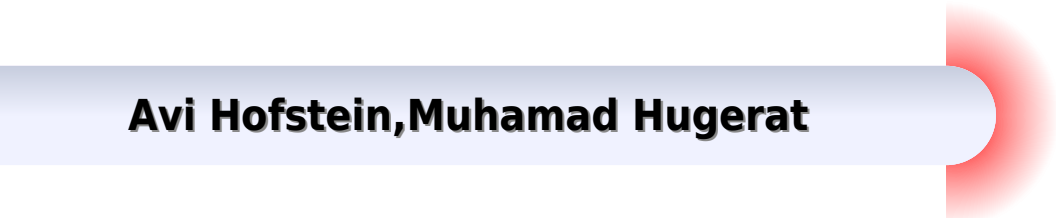


- d. Write the equation for the dehydration of the CoCl_2 hydrate.



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Avi Hofstein, Muhamad Hugerat



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Chemistry Neil D. Jespersen, Alison Hyslop, 2021-11-02 Chemistry The Molecular Nature of Matter 8th Edition continues to focus on the intimate relationship that exists between structure at the atomic molecular level and the observable macroscopic properties of matter Key revisions in this edition focus on three areas The deliberate inclusion of more updated real world examples that relate common real world student experiences to the science of chemistry Simultaneously examples and questions have been updated to align them with career concepts relevant to the environmental engineering biological pharmaceutical and medical sciences Providing students with transferable skills with a focus on integrating metacognition and three dimensional learning into the text When students know what they know they are better able to learn and incorporate the material Providing a total solution through New WileyPLUS by fully integrating the enhanced etext with online assessment answer specific responses and additional practice resources The 8th edition continues to emphasize the importance of applying concepts to problem solving to achieve high level learning and increase retention of chemistry knowledge Problems are arranged in an intuitive confidence building order **Practical Chemistry** Teshome

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