

Periodic Trends Lab

Periodic law states that when the elements are arranged in order of increasing atomic number, there is a periodic repetition of their chemical and physical properties. Elements within the same group have the most similar chemical and physical properties, although not identical. Elements in neighboring groups may also have similar chemical and physical properties, but to a lesser degree. Trends in properties occur within groups and across periods.

PURPOSE: In this activity, you will

- Interpret group and periodic trends in atomic radius, electronegativity, and ionization energy.
- Graph these trends.

PROCEDURE:

1. Complete the pre-lab section.
2. Using the data from **Table S** of your reference tables, prepare the graphs listed below. Staple your graphs to the back of this lab packet.

On each graph be sure to:

- **USE PENCIL!**
- Put an appropriate title and labels for each axis.
- Omit the transition elements.
- Connect the points you plot.
- Connect the points representing Group 1 elements with a green line.
- Connect the points representing Group 2 elements with a blue line.
- Connect the points representing Group 17 elements with a red line.
- Draw a triangle around the points representing the elements of Period 3.

GRAPH 1: Atomic radius vs. atomic number for elements 1-54.

GRAPH 2: Electronegativity vs. atomic number for elements 1-54.

GRAPH 3: Ionization energy vs. atomic number for elements 1-54.

3. Complete the conclusion section.

Chemistry Periodic Trends Lab Activity

Terry C. Jones



Chemistry Periodic Trends Lab Activity:

Exploring General Chemistry in the Laboratory Colleen F. Craig, Kim N. Gunnerson, 2017-02-01 This laboratory manual is intended for a two semester general chemistry course The procedures are written with the goal of simplifying a complicated and often challenging subject for students by applying concepts to everyday life This lab manual covers topics such as composition of compounds reactivity stoichiometry limiting reactants gas laws calorimetry periodic trends molecular structure spectroscopy kinetics equilibria thermodynamics electrochemistry intermolecular forces solutions and coordination complexes By the end of this course you should have a solid understanding of the basic concepts of chemistry which will give you confidence as you embark on your career in science Lab Experiments in Introductory Chemistry Phil Reedy, Donald J.

Wink, Sharon Fetzer-Gislason, 2003-03-21 The manual contains laboratory experiments written specifically for the prep chem lab as well as for the general chemistry course Available as a complete manual or custom published at <http://custompub.whfreeman.com> Chemistry McGraw-Hill Staff, 2001-07

Hands-On General Science Activities With Real-Life Applications Pam Walker, Elaine Wood, 2008-04-21 In this second edition of Hands On General Science Activities with Real Life Applications Pam Walker and Elaine Wood have completely revised and updated their must have resource for science teachers of grades 5 12 The book offers a dynamic collection of classroom ready lessons projects and lab activities that encourage students to integrate basic science concepts and skills into everyday life **Principles Of Descriptive**

Inorganic Chemistry Gary Wulfsberg, 1991-05-29 This is the only text currently available organized by class of compound and by property or reaction type not group by group or element by element which requires students to memorize isolated facts This is the only text currently available organized by class of compound and by property or reaction type not group by group or element by element which requires students to memorize isolated facts Translated into Italian **Chemistry in**

the Community. American Chemical Society, 2002 This volume has relevance to a wide number of courses giving a hands on introduction to chemistry in relation to community issues rather than around specific chemical concepts Holt Chemistry Ralph Thomas Myers, 2004 **Prentice Hall Chemistry**, 2000 **Innovations in Chemistry Education: Strategies for**

Engaged and Effective Learning Dr. S. Vincent De Paul, 2024-11-14 *Innovations in Chemistry Education Strategies for Engaged and Effective Learning* is a comprehensive guide that explores innovative strategies for improving chemistry education at all levels The book addresses key challenges faced by chemistry educators today and offers effective solutions to engage students and promote deeper learning The book is organized into thematic sections that cover a range of topics including active learning techniques the integration of technology such as virtual labs and simulations and inclusive teaching practices It discusses the use of contemporary tools such as AI and machine learning to personalize learning experiences as well as strategies for fostering collaboration and critical thinking Each chapter presents actionable strategies and examples making this book a valuable resource for educators who seek to transform their teaching methods and provide students with

a more interactive and engaging learning experience Designed for both experienced and new educators as well as students and researchers in the field of chemistry education this book provides insights into how to adapt chemistry teaching to modern needs It offers a vision for a more inclusive technology driven and student centered approach to teaching chemistry that prepares learners for the challenges of tomorrow s scientific landscape Who's the New Kid in Chemistry? John D. Butler, 2013-12-12 Who s the New Kid in Chemistry offers an unprecedented look at student engagement and teacher best practices through the eyes of an educational researcher enrolled as a public high school student Over the course of seventy nine consecutive days John D Butler participates in and observes Rhode Island 2013 Teacher of the Year Jessica M Waters s high school chemistry class documenting his experiences as they unfold Who s the New Kid in Chemistry is a compelling example of what can be accomplished when an educational researcher and teacher collaborate in the classroom This work includes a discussion on flexible homework assignments data driven instruction and thirty teacher best practices This book is an invaluable resource for teachers across all content areas masters and doctoral research method classes and future Teachers of the Year

Unveiling the Magic of Words: A Review of "**Chemistry Periodic Trends Lab Activity**"

In a global defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their capability to kindle emotions, provoke contemplation, and ignite transformative change is truly awe-inspiring. Enter the realm of "**Chemistry Periodic Trends Lab Activity**," a mesmerizing literary masterpiece penned by way of a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve in to the book is central themes, examine its distinctive writing style, and assess its profound effect on the souls of its readers.

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