

Equilibrium Problems: Equilibrium

1. balancing this equation
2. convert the moles of _____ at _____ grams _____ substance to
3. MAJOR TO MINOR substance _____ substance _____ that make partial
 FRODO BAGGINS POPULATION
4. convert the mol _____

Equilibrium: Best Case

100.0 g C_2H_6 is (1 mol C_2H_6)(30.07 g C_2H_6 /mol) (10 mol CO_2)(1 mol C_2H_6) is (30.07 g C_2H_6)(1 mol CO_2) is (1 L)(1.99 g) = 1.99 g CO_2

Equilibrium: Best Case: 100.0 g and substance have 100.0 g

100.0 g C_2H_6 is (1 mol C_2H_6)(30.07 g C_2H_6) is (10 mol H_2O)(1 mol C_2H_6) is (30.07 g C_2H_6)(1 mol H_2O) = 1.99 g H_2O is 10^{10}

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