

Binary Ionic Compounds

- Made up of _____ cations _____ anions.
- Name the _____ cation _____ anion _____.
- Name the _____ cation _____ anion _____, a charged ion, adding 1 to _____.
- Always include their ionic forms _____ that are fixed and _____ and (as is implied, not as _____) in parentheses after the name of the _____ in case it is _____.

Examples: NaCl sodium chloride
 Cu_2S copper sulfide

Hg_2Cl_2 mercurous chloride
 SnCl_2 tin (II) chloride

Binary Acid Compounds

- Made up of _____ cations _____ anions.
- Name the _____ that name the _____ polyatomic anion _____ will have changing the ending to "ide".

Examples: HNO_3 nitric acid
 HClO_4 perchloric acid

Chemistry Note Taking Guide Episode 603

William Swan Sonnenschein



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