THE "OFFICIAL" CHEMISTRY 12 REDOX &ELECTROCHEMISTRY STUDY GUIDE

- DO ALL THE QUESTIONS in this booklet. These are actual Provincial Exam questions? Your own provincial exam and unit test will include questions similar to the ones in this booklet?
- RESIST THE URGE TO LOOK AT THE ANSWER KEY until you have given all the questions in the section your best effort. Don't do
 one question, then look at the key, then do another and look at the key, and so on. Each time you look at one answer in the study
 quete, your eye will notice other answers around them, and this will reduce the effectiveness of trose questions in helping you to learn.
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- Check Off the STATUS box on the PRESCRIBED LEARNING OUTCOMES sheet. I have tried to organize the questions in the identical sequence to which they appear on your Redoor/Electrochemistry Prescribed Learning Outcome sheet. By doing this, you can be confident that you know everything you need to know for both the UNIT EXAM and PROVINCIAL EXAM!

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INTRODUCTION

- 1. set Which one of the following is the strongest reducing agent?
- 2. Set In the reaction below:
 - $6H^+ + 6I^- + C1O_3^- \rightarrow 3I_2 + 3H_2O + CI^-$
- A. 12 B. F C. H* (5) CIO 3.
- 3. 501 Which of the following is the strongest reducing agent?
 - A. cobult
 - B. copper calcium
- 4. 501 An oxidizing agent is

cheomium

- A. reduced as it loses electrons.
 reduced as it gains electrons.
 oxidized as it loses electrons.
 D. oxidized as it gains electrons.
- 5. 501 Which of the following is the strongest exidizing agent?

B. Mn²⁺
C acidified MnO₂
D acidified MnO₃

Chemistry Electrochemistry Study Guide Answers Key

Arshad Iqbal

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