

Academic Chemistry 2nd Semester Final 2022

Thermodynamics

13. The average kinetic energy of the particles in solids, liquids, and gases is measured as?

- a. Temperature
- b. Density
- c. Volume
- d. Pressure

Use the graph below to answer questions 14-16.



14. What phase change is occurring at 4?

- a. melting
- b. freezing
- c. vaporization
- d. none of the above

15. Is this diagram representing an exothermic or endothermic reaction?

- a. exothermic
- b. endothermic

16. Which stage of the graph would you find water as both a liquid and a solid?

- a. 2
- b. 3
- c. 4
- d. 5

17. During a phase change, the temperature:

- a. increases
- b. stays the same
- c. decreases

18. A sample of gold absorbs 50.6 Joules, causing its temperature to rise from 25.0°C to 82.7°C . If the specific heat capacity of gold is $0.130 \text{ J/g}^{\circ}\text{C}$, what is the mass of the sample?

- a. 0.148 grams
- b. 8.16 grams
- c. 6.75 grams
- d. 22.2 grams

Chemistry 2nd Semester Finals

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Table of Contents Chemistry 2nd Semester Finals

1. Understanding the eBook Chemistry 2nd Semester Finals
 - The Rise of Digital Reading Chemistry 2nd Semester Finals
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry 2nd Semester Finals
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry 2nd Semester Finals
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry 2nd Semester Finals
 - Personalized Recommendations
 - Chemistry 2nd Semester Finals User Reviews and Ratings
 - Chemistry 2nd Semester Finals and Bestseller Lists

5. Accessing Chemistry 2nd Semester Finals Free and Paid eBooks
 - Chemistry 2nd Semester Finals Public Domain eBooks
 - Chemistry 2nd Semester Finals eBook Subscription Services
 - Chemistry 2nd Semester Finals Budget-Friendly Options
6. Navigating Chemistry 2nd Semester Finals eBook Formats
 - ePub, PDF, MOBI, and More
 - Chemistry 2nd Semester Finals Compatibility with Devices
 - Chemistry 2nd Semester Finals Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemistry 2nd Semester Finals
 - Highlighting and Note-Taking Chemistry 2nd Semester Finals
 - Interactive Elements Chemistry 2nd Semester Finals
8. Staying Engaged with Chemistry 2nd Semester Finals
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry 2nd Semester Finals
9. Balancing eBooks and Physical Books Chemistry 2nd Semester Finals
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry 2nd Semester Finals
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Chemistry 2nd Semester Finals
 - Setting Reading Goals Chemistry 2nd Semester Finals
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chemistry 2nd Semester Finals
 - Fact-Checking eBook Content of Chemistry 2nd Semester Finals
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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