Chemistry 11

Some Study Materials for the Final Exam

Prefix	Abbreviation	Exponent
giga	G	10"
mega	M	104
kilo	k	103
hecto	h	102
deca	da	101
deci	d	10'1
centi	c	10-2
milli	m	10-3
micro	ja:	10-6
nano	n n	10-7
pico	p	10.12

Complex Unit Conversion

Density



Precision

- The number of significant digits to which a value has been reliably measured.
- -The more decimal places, the more precise the measurement.
- -The greater the precision, the less the uncertainty.

The accuracy of a measurement is how close the measurement is to an accepted standard.

Significant Digits Rules:

- 1. All non-zero digits are significant
- 2. Zero's between non-zero digits are significant
- 3. All zero's to the LEFT of the 1st non-zero digit are NOT significant. (called leading zero's)
- 4. Zero's to the RIGHT of the last non-zero digit ARE significant IF the decimal point is SHOWN.
- Zero's to the RIGHT of the last non-zero digit (trailing zero's) ARE NOT significant if there is an UNDERSTOOD decimal point.
- In Scientific Notation, the exponent part of the number is NOT significant.

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IM Harris

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