

Chemistry Worksheet Class 11 on Chapter 1 Some Basic Concepts of Chemistry - Set 4

Q1. Calculate the correct number of significant figures: 4.26 - (15.635/5.0)

a.) 1.16

b.) 1.12

c.) 1.2

d.) 1.133

Q2. What amount of dioxygen (in gram) contains 1.8 × 10²² molecules?

a.) 0.960

b.) 96.0

c.) 0.0960

d.) 9.60

Q3. If we consider % in lace of 1/12 mass of carbon atom is taken to be the relative atomic mass unit, the mass of one mole of a substance will—

a.) Decrease twice

b.) Increase two folds

c.) Remains unchanged

d.) Be a fraction of the molecular mass of the substance

Q4. The molarity of a solution obtained by mixing 800 mL of 0.5 M HCl with 200 mL of 1 M HCl will be

a.) 0.8 M

b.) 0.6 M

c.) 0.4 M

d.) 0.2 M

Q5. The number of molecules present in 1 ml of water is ____.

a.) 1

b.) 1000

c.) 2.69 × 10¹⁹

d.) 6.02×10^{20}

Q6. Define the atomic mass of an element.

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