

Chemical Reactions

Section 9.1 Reactions and Equations

In your textbook, read about evidence of chemical reactions.

For each statement, write *yes* if evidence of a chemical reaction is present. Write *no* if there is no evidence of a chemical reaction.

- | | |
|---------------|---|
| <u>yes</u> | 1. A tomato smells rotten. |
| <u>no</u> | 2. A drinking glass breaks into smaller pieces. |
| <u>no</u> | 3. A piece of ice melts. |
| <u>yes</u> | 4. Drain cleaner is mixed with water and the solution becomes warm. |
| <u>yes</u> | 5. Candle wax burns. |
| <u>yes/no</u> | 6. Molten candle wax solidifies. |
| <u>yes</u> | 7. Green leaves turn yellow and red as the seasons change. |
| <u>yes</u> | 8. Baking powder produces a gas that makes a cake rise. |

In your textbook, read about how to represent chemical reactions and how to balance chemical equations.

Use the terms below to complete the passage. Each term may be used once, more than once, or not at all.

arrow	plus sign	(s)	(l)
reactant	product	(g)	(aq)

The fuel for the space shuttle is hydrogen, which burns in oxygen to produce water vapor and energy. In this chemical reaction, hydrogen is a(n) **(9)** reactant, oxygen is a(n) **(10)** reactant, and water vapor is a(n) **(11)** product. In a chemical equation for this reaction, a(n) **(12)** arrow is used to separate hydrogen and oxygen from water vapor and energy. A(n) **(13)** plus sign is used to separate the symbols for hydrogen and oxygen. A(n) **(14)** (g) symbol is used to tell the state of hydrogen in the reaction, a(n) **(15)** (g) symbol is used for the state of oxygen, and a(n) **(16)** (g) symbol is used for the state of water vapor.

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Table of Contents Chemical Reactions Study Guide Packet Answers

1. Understanding the eBook Chemical Reactions Study Guide Packet Answers
 - The Rise of Digital Reading Chemical Reactions Study Guide Packet Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemical Reactions Study Guide Packet Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemical Reactions Study Guide Packet Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemical Reactions Study Guide Packet Answers
 - Personalized Recommendations
 - Chemical Reactions Study Guide Packet Answers User Reviews and Ratings
 - Chemical Reactions Study Guide Packet Answers and Bestseller Lists

5. Accessing Chemical Reactions Study Guide Packet Answers Free and Paid eBooks
 - Chemical Reactions Study Guide Packet Answers Public Domain eBooks
 - Chemical Reactions Study Guide Packet Answers eBook Subscription Services
 - Chemical Reactions Study Guide Packet Answers Budget-Friendly Options
6. Navigating Chemical Reactions Study Guide Packet Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Chemical Reactions Study Guide Packet Answers Compatibility with Devices
 - Chemical Reactions Study Guide Packet Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemical Reactions Study Guide Packet Answers
 - Highlighting and Note-Taking Chemical Reactions Study Guide Packet Answers
 - Interactive Elements Chemical Reactions Study Guide Packet Answers
8. Staying Engaged with Chemical Reactions Study Guide Packet Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemical Reactions Study Guide Packet Answers
9. Balancing eBooks and Physical Books Chemical Reactions Study Guide Packet Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemical Reactions Study Guide Packet Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Chemical Reactions Study Guide Packet Answers
 - Setting Reading Goals Chemical Reactions Study Guide Packet Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chemical Reactions Study Guide Packet Answers
 - Fact-Checking eBook Content of Chemical Reactions Study Guide Packet Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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