

Post-Lab Questions

1. From the experiment, which reactant is the limiting reactant? Which one is the reactant in excess? How do you determine which one is the limiting reactant?
2. If you weigh the watch glass then write your name on it with the wax pencil, how will that affect your percent yield?
3. What factors could cause the percent yield of a product to be less than 100%?
4. What factors could cause the percent yield of a product to be more than 100%?
5. Balance the following reaction: $\text{C}_3\text{H}_8 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$
 - a. 4.30 g of C_3H_8 reacts with 4.30 g of O_2 . Which one is the limiting reagent, and which one is in excess? Explain your reasoning.
 - b. Based on the limiting reagent, what are the theoretical yields of CO_2 and H_2O ?
 - c. You perform the experiment and produce 2.67 g of CO_2 and 0.79 g of H_2O . What are your percent yields for each product?

Chem 1211 Lab Manual

Msu Chemistry Department



Chem 1211 Lab Manual:

Unveiling the Power of Verbal Artistry: An Psychological Sojourn through **Chem 1211 Lab Manual**

In a global inundated with displays and the cacophony of immediate conversation, the profound energy and mental resonance of verbal artistry often fade in to obscurity, eclipsed by the constant barrage of sound and distractions. Yet, located within the musical pages of **Chem 1211 Lab Manual**, a charming perform of fictional elegance that impulses with natural thoughts, lies an memorable journey waiting to be embarked upon. Published by a virtuoso wordsmith, this interesting opus books viewers on a mental odyssey, lightly revealing the latent possible and profound influence embedded within the delicate web of language. Within the heart-wrenching expanse with this evocative analysis, we can embark upon an introspective exploration of the book is key subjects, dissect its interesting writing design, and immerse ourselves in the indelible impact it leaves upon the depths of readers souls.

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