

G. C. 2-9: LABORATORY REPORT

Name: Kassidy Heinz

YOU MUST SHOW ALL YOUR CALCULATIONS TO RECEIVE CREDIT FOR THEM!

Part I. Addition of 0.100 M HCl to Buffer and Deionized Water

Concentration of buffer: For acetic acid: 0.1 M For acetate: 0.1 M

Volume of HCl	Total Volume of HCl	pH Meter Reading		Expected pH (Calculated)	
		Buffer	Water	Buffer	Water
0	0	4.76	7.00	4.76	
5.00	5.00	4.67	2.04		
5.00	10.00	4.58	1.78		
10.00	20.00	4.39	1.54		
10.00	30.00	4.16	1.43		
20.00	50.00	2.88	1.30		
50.00	100.00	1.48	1.18		

Calculations must be shown in your laboratory notebook.

Part II. Addition of 0.100 M NaOH to Buffer and Deionized water

Volume of NaOH	Total Volume NaOH	pH Meter Reading		Expected pH (Calculated)	
		Buffer	Water	Buffer	Water
0	0	4.76	7		
5.00	5.00	4.84	11.95		
5.00	10.00	4.93	12.22		
10.55	20.00	5.13	12.45		
10.00	30.00	5.36	12.52		
20.00	50.00	8.88	12.69		
50.00	100.00	12.51	12.82		

Calculations must be shown in your laboratory notebook.

Chm 2046 Lab Manual

Jicheng Xie



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Table of Contents Chm 2046 Lab Manual

1. Understanding the eBook Chm 2046 Lab Manual
 - The Rise of Digital Reading Chm 2046 Lab Manual
 - Advantages of eBooks Over Traditional Books
2. Identifying Chm 2046 Lab Manual
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chm 2046 Lab Manual
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chm 2046 Lab Manual
 - Personalized Recommendations
 - Chm 2046 Lab Manual User Reviews and Ratings

- Chm 2046 Lab Manual and Bestseller Lists
- 5. Accessing Chm 2046 Lab Manual Free and Paid eBooks
 - Chm 2046 Lab Manual Public Domain eBooks
 - Chm 2046 Lab Manual eBook Subscription Services
 - Chm 2046 Lab Manual Budget-Friendly Options
- 6. Navigating Chm 2046 Lab Manual eBook Formats
 - ePub, PDF, MOBI, and More
 - Chm 2046 Lab Manual Compatibility with Devices
 - Chm 2046 Lab Manual Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chm 2046 Lab Manual
 - Highlighting and Note-Taking Chm 2046 Lab Manual
 - Interactive Elements Chm 2046 Lab Manual
- 8. Staying Engaged with Chm 2046 Lab Manual
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chm 2046 Lab Manual
- 9. Balancing eBooks and Physical Books Chm 2046 Lab Manual
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chm 2046 Lab Manual
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Chm 2046 Lab Manual
 - Setting Reading Goals Chm 2046 Lab Manual
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chm 2046 Lab Manual
 - Fact-Checking eBook Content of Chm 2046 Lab Manual
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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