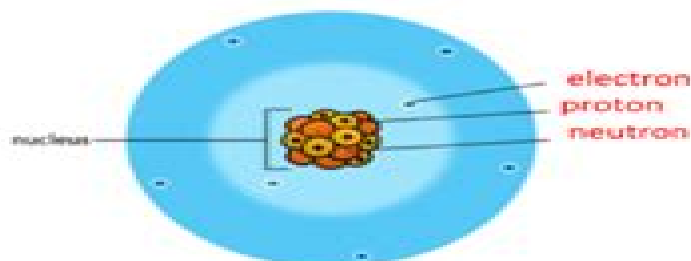


# CHEMISTRY TEST STUDY GUIDE ANSWER KEY

Complete the following information about atoms.  
Label the subatomic particles in the atom diagram below.



What **charge** does each subatomic particle have?

Proton: **positive**

Electron: **negative**

Neutron: **neutral (no charge)**

What subatomic particles are found in the **nucleus**?

**protons and neutrons**

Classify each of the following substances as an Element (E), Compound (C), or Mixture (M).

*If it's a mixture, is it heterogeneous (he) or homogeneous (ho)?*

- water **C**
- salt **C**
- air **M (ho)**
- carbon **E**
- sugar **C**
- saltwater **M (ho)**
- hydrogen **E**
- carbon dioxide **C**
- kool aid **M (ho)**

Use a periodic table to answer the following questions.

- Is sodium a metal or nonmetal? **metal**
- What is the atomic number of oxygen? **8**
- What is element symbol for iron? **Fe**
- What other elements are in the same family as Chlorine? **Fluorine, Bromine, Iodine, Astatine, Tennessine**
- Li, Be, B, C, N, O, F, and Ne are all in the same: **Period**

# Ch21 Study Guide Answer Key For Chemistry

**Neil E. Schore**



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with one largely based on reaction mechanisms The authors philosophy is to emphasize mechanisms and their common aspects as often as possible and at the same time use the unifying features of functional groups as the basis for most chapters The structural aspects of the authors approach show students what organic chemistry is Mechanistic aspects of their approach show students how it works And wherever an opportunity arises the authors show students what it does in living systems and the physical world around us

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