

## Names & Formulas for Ionic Compounds

Give the name or formula of the following ionic compounds:

| Name |                                            | Formula |                              |
|------|--------------------------------------------|---------|------------------------------|
| 1)   | $\text{Na}_2\text{CO}_3$ _____             | 21)     | sodium phosphide _____       |
| 2)   | $\text{NaOH}$ _____                        | 22)     | magnesium nitrate _____      |
| 3)   | $\text{MgBr}_2$ _____                      | 23)     | lead (II) sulfite _____      |
| 4)   | $\text{KCl}$ _____                         | 24)     | calcium phosphate _____      |
| 5)   | $\text{FeCl}_2$ _____                      | 25)     | ammonium sulfate _____       |
| 6)   | $\text{FeCl}_3$ _____                      | 26)     | silver cyanide _____         |
| 7)   | $\text{Zn}(\text{OH})_2$ _____             | 27)     | aluminum sulfide _____       |
| 8)   | $\text{Be}_2\text{SO}_4$ _____             | 28)     | beryllium chloride _____     |
| 9)   | $\text{CrF}_2$ _____                       | 29)     | copper (I) arsenide _____    |
| 10)  | $\text{Al}_2\text{S}_3$ _____              | 30)     | iron (III) oxide _____       |
| 11)  | $\text{PbO}$ _____                         | 31)     | gallium nitride _____        |
| 12)  | $\text{Li}_3\text{PO}_4$ _____             | 32)     | iron (II) bromide _____      |
| 13)  | $\text{TiI}_4$ _____                       | 33)     | vanadium (V) phosphate _____ |
| 14)  | $\text{Co}_3\text{N}_2$ _____              | 34)     | calcium oxide _____          |
| 15)  | $\text{Mg}_3\text{P}_2$ _____              | 35)     | magnesium acetate _____      |
| 16)  | $\text{Ga}(\text{NO}_2)_3$ _____           | 36)     | aluminum sulfate _____       |
| 17)  | $\text{Ag}_2\text{SO}_3$ _____             | 37)     | copper (I) carbonate _____   |
| 18)  | $\text{NH}_4\text{OH}$ _____               | 38)     | barium oxide _____           |
| 19)  | $\text{Al}(\text{CN})_3$ _____             | 39)     | ammonium sulfite _____       |
| 20)  | $\text{Be}(\text{CH}_3\text{COO})_2$ _____ | 40)     | silver bromide _____         |

# Compound Names And Formulas Activity Answer Key

**Yicheng Fang**



## Compound Names And Formulas Activity Answer Key:

*Physical and Chemical Changes* Edward P. Ortleb, Richard Cadice, 1993-09-01 General chemistry information including everything from matter to radioactivity For grades 5 to 9      **Prentice Hall Chemistry**, 2000      Inorganic Chemistry Tina Overton, Jonathan Rourke, Fraser A. Armstrong, 2018 Leading the reader from the fundamental principles of inorganic chemistry right through to cutting edge research at the forefront of the subject Inorganic Chemistry Seventh Edition is the ideal course companion for the duration of a student's degree The authors have drawn upon their extensive teaching and research experience to update this text the seventh edition retains the much praised clarity of style and layout from previous editions while offering an enhanced section on expanding our horizons The latest innovative applications of green chemistry have been added to clearly illustrate the real world significance of the subject This edition also sees a greater use of learning features including substantial updates to the problem solving questions additional self tests and walk through explanations which enable students to check their understanding of key concepts and develop problem solving skills Providing comprehensive coverage of inorganic chemistry while placing it in context this text will enable the reader to fully master this important subject Online Resources Inorganic Chemistry Seventh Edition is accompanied by a range of online resources For registered adopters of the text DT Figures marginal structures and tables of data ready to download DT Test bank For students DT Answers to self tests and exercises from the book DT Tables for group theory DT Web links DT Links to interactive structures and other resources on [www.chemtube3d.com](http://www.chemtube3d.com)      *Prentice Hall Science Explorer: Chemical Interactions* Pearson/Prentice Hall, Michael J. Padilla, 2004-10      **Popular Science**, 2005-09 Popular Science gives our readers the information and tools to improve their technology and their world The core belief that Popular Science and our readers share The future is going to be better and science and technology are the driving forces that will help make it better

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## **Compound Names And Formulas Activity Answer Key** Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Compound Names And Formulas Activity Answer Key**," compiled by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we shall delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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web figure pageindex 1 types of root systems in the figure above label any adventitious roots prop roots and storage roots

label each system as either netted or taproot except the topmost root system which is an underground stem

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