Chemical Reactions

Signs of Chemical Reactions

What is a chemical reaction?

Chemical Formulas

H₂O

A chemical formula uses ______ to identify the ______ in a correpound and the

of each kind present in a molecule.

C₆H₁₂O₆

A glucese molecule is made up of _____ atoms
of _____ atoms of

Chemical Equations

* attionis

charmical equation.

chemical formula

chemical reaction.

· compound

A chemical equation uses ________ to describe a

chemical reaction. The starting materials in a chemical reaction are
_____. The resulting substances from a reaction

placed on the left of the arrow, and the ______ on the right of the arrow.

 $C + O_2 \rightarrow CO_2$

Types of Chemical Reactions

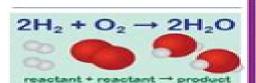
Synthesis Reaction

Describe a synthesis reaction:

A warter mobile-pulse is made

bonded to _____ atom of

up-of _____ atoms of





These simple shapes combine to form a new object.

Decomposition Reaction

attorns of

Describe a decomposition reaction:

0116		211		*
2H ₂ C	, —		+ 0	12
		-0		9
-		.00	-	
reactas	st pro	iduct +	produc	A.
	The same of the sa			
		4		
	~	The first	oth return	
		been all a	Character Senter	

Replacement Reaction

double replacement reaction.

single replacement reaction

Single Replacement

decomposition reaction - swithesis reaction.

Describe a single replacement reaction:

Key Vocabulary Terms

* effectivene

* controlling to

* PRINCESOR

Zn + 2HCl → ZnCl₂ + H₂ reactant + reactant → product + product



Double Replacement

Describe à double replacement reaction:

NaCl + AgF → NaF + AgCl reactant + reactant → product + product



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Chemical Reactions Guide

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