

# THE PERIODIC TABLE OF ELEMENTS

| THE PERIODIC TABLE OF ELEMENTS |                                                          |                                                           |                                                          |                                                                |                                                          |                                                               |                                                            |                                                            |                                                             |                                                               |                                                              |                                                              |                                                              |                                                              |                                                            |                                                              |                                                             |                                                             |                                                       |                                                       |
|--------------------------------|----------------------------------------------------------|-----------------------------------------------------------|----------------------------------------------------------|----------------------------------------------------------------|----------------------------------------------------------|---------------------------------------------------------------|------------------------------------------------------------|------------------------------------------------------------|-------------------------------------------------------------|---------------------------------------------------------------|--------------------------------------------------------------|--------------------------------------------------------------|--------------------------------------------------------------|--------------------------------------------------------------|------------------------------------------------------------|--------------------------------------------------------------|-------------------------------------------------------------|-------------------------------------------------------------|-------------------------------------------------------|-------------------------------------------------------|
| 1                              |                                                          |                                                           |                                                          |                                                                |                                                          |                                                               |                                                            |                                                            |                                                             |                                                               |                                                              |                                                              |                                                              |                                                              |                                                            |                                                              | 18                                                          |                                                             |                                                       |                                                       |
| 1                              | <div><div>1</div><div>H</div><div>Hydrogen</div></div>   |                                                           |                                                          |                                                                |                                                          |                                                               |                                                            |                                                            |                                                             |                                                               |                                                              |                                                              |                                                              |                                                              |                                                            |                                                              |                                                             | 2                                                           | <div><div>2</div><div>He</div><div>Helium</div></div> |                                                       |
| 2                              | <div><div>3</div><div>Li</div><div>Lithium</div></div>   | <div><div>4</div><div>Be</div><div>Beryllium</div></div>  |                                                          |                                                                |                                                          |                                                               |                                                            |                                                            |                                                             |                                                               |                                                              |                                                              |                                                              |                                                              |                                                            |                                                              |                                                             |                                                             | 10                                                    | <div><div>10</div><div>Ne</div><div>Neon</div></div>  |
| 3                              | <div><div>11</div><div>Na</div><div>Sodium</div></div>   | <div><div>12</div><div>Mg</div><div>Magnesium</div></div> |                                                          |                                                                |                                                          |                                                               |                                                            |                                                            |                                                             |                                                               |                                                              |                                                              |                                                              |                                                              |                                                            |                                                              |                                                             |                                                             | 18                                                    | <div><div>18</div><div>Ar</div><div>Argon</div></div> |
| 4                              | <div><div>19</div><div>K</div><div>Potassium</div></div> | <div><div>20</div><div>Ca</div><div>Calcium</div></div>   | <div><div>21</div><div>Sc</div><div>Scandium</div></div> | <div><div>22</div><div>Ti</div><div>Titanium</div></div>       | <div><div>23</div><div>V</div><div>Vanadium</div></div>  | <div><div>24</div><div>Cr</div><div>Chromium</div></div>      | <div><div>25</div><div>Mn</div><div>Manganese</div></div>  | <div><div>26</div><div>Fe</div><div>Iron</div></div>       | <div><div>27</div><div>Co</div><div>Cobalt</div></div>      | <div><div>28</div><div>Ni</div><div>Nickel</div></div>        | <div><div>29</div><div>Cu</div><div>Copper</div></div>       | <div><div>30</div><div>Zn</div><div>Zinc</div></div>         | <div><div>31</div><div>Ga</div><div>Gallium</div></div>      | <div><div>32</div><div>Ge</div><div>Germanium</div></div>    | <div><div>33</div><div>As</div><div>Arsenic</div></div>    | <div><div>34</div><div>Se</div><div>Selenium</div></div>     | <div><div>35</div><div>Br</div><div>Bromine</div></div>     | <div><div>36</div><div>Kr</div><div>Krypton</div></div>     |                                                       |                                                       |
| 5                              | <div><div>37</div><div>Rb</div><div>Rubidium</div></div> | <div><div>38</div><div>Sr</div><div>Strontium</div></div> | <div><div>39</div><div>Y</div><div>Yttrium</div></div>   | <div><div>40</div><div>Zr</div><div>Zirconium</div></div>      | <div><div>41</div><div>Nb</div><div>Niobium</div></div>  | <div><div>42</div><div>Mo</div><div>Molybdenum</div></div>    | <div><div>43</div><div>Tc</div><div>Technetium</div></div> | <div><div>44</div><div>Ru</div><div>Ruthenium</div></div>  | <div><div>45</div><div>Rh</div><div>Rhodium</div></div>     | <div><div>46</div><div>Pd</div><div>Palladium</div></div>     | <div><div>47</div><div>Ag</div><div>Silver</div></div>       | <div><div>48</div><div>Cd</div><div>Cadmium</div></div>      | <div><div>49</div><div>In</div><div>Indium</div></div>       | <div><div>50</div><div>Sn</div><div>Tin</div></div>          | <div><div>51</div><div>Sb</div><div>Antimony</div></div>   | <div><div>52</div><div>Te</div><div>Tellurium</div></div>    | <div><div>53</div><div>I</div><div>Iodine</div></div>       | <div><div>54</div><div>Xe</div><div>Xenon</div></div>       |                                                       |                                                       |
| 6                              | <div><div>55</div><div>Cs</div><div>Cesium</div></div>   | <div><div>56</div><div>Ba</div><div>Barium</div></div>    | <div><div>57-71</div><div>Lanthanide Series</div></div>  | <div><div>72</div><div>Hf</div><div>Hafnium</div></div>        | <div><div>73</div><div>Ta</div><div>Tantalum</div></div> | <div><div>74</div><div>W</div><div>Tungsten</div></div>       | <div><div>75</div><div>Re</div><div>Rhenium</div></div>    | <div><div>76</div><div>Os</div><div>Osmium</div></div>     | <div><div>77</div><div>Ir</div><div>Iridium</div></div>     | <div><div>78</div><div>Pt</div><div>Platinum</div></div>      | <div><div>79</div><div>Au</div><div>Gold</div></div>         | <div><div>80</div><div>Hg</div><div>Mercury</div></div>      | <div><div>81</div><div>Tl</div><div>Thallium</div></div>     | <div><div>82</div><div>Pb</div><div>Lead</div></div>         | <div><div>83</div><div>Bi</div><div>Bismuth</div></div>    | <div><div>84</div><div>Po</div><div>Polonium</div></div>     | <div><div>85</div><div>At</div><div>Astatine</div></div>    | <div><div>86</div><div>Rn</div><div>Radon</div></div>       |                                                       |                                                       |
| 7                              | <div><div>87</div><div>Fr</div><div>Francium</div></div> | <div><div>88</div><div>Ra</div><div>Radium</div></div>    | <div><div>89-103</div><div>Actinide Series</div></div>   | <div><div>104</div><div>Rf</div><div>Rutherfordium</div></div> | <div><div>105</div><div>Db</div><div>Dubnium</div></div> | <div><div>106</div><div>Sg</div><div>Seaborgium</div></div>   | <div><div>107</div><div>Bh</div><div>Bohrium</div></div>   | <div><div>108</div><div>Hs</div><div>Hassium</div></div>   | <div><div>109</div><div>Mt</div><div>Meitnerium</div></div> | <div><div>110</div><div>Ds</div><div>Darmstadtium</div></div> | <div><div>111</div><div>Rg</div><div>Roentgenium</div></div> | <div><div>112</div><div>Cn</div><div>Copernicium</div></div> | <div><div>113</div><div>Nh</div><div>Nihonium</div></div>    | <div><div>114</div><div>Fl</div><div>Flerovium</div></div>   | <div><div>115</div><div>Mc</div><div>Moscovium</div></div> | <div><div>116</div><div>Lv</div><div>Livermorium</div></div> | <div><div>117</div><div>Ts</div><div>Tennessine</div></div> | <div><div>118</div><div>Og</div><div>Oganesson</div></div>  |                                                       |                                                       |
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Atomic Number → 1  
Symbol → H  
Name → Hydrogen  
Average Atomic Mass → 1.008

- Alkali Metals
- Alkaline Earth Metals
- Transition Metals
- Other Metals
- Metalloids
- Non-metals
- Halogens
- Noble Gases
- Lanthanides
- Actinides

# Dimstri Chemistry Of Elements

**Torsten Schmiermund**



## **Dimstri Chemistry Of Elements:**

Dimstri Chemistry of Elements P. Subarat, 2015-05-18 New Researched Findings In Field of Chemical Sciences

**30-Second Elements** Eric Scerri, 2013-08-01 When was radium discovered Who are Dmitri Mendeleev and Glenn T Seaborg Who discovered uranium's radioactivity Which element is useful for dating the age of Earth And why doesn't gold have a scientific name 30 Second Elements presents you with the very foundations of chemical knowledge explaining concisely the 50 most significant chemical elements This book uses helpful glossaries and tables to fast track your knowledge of the other 68 elements and the relationships between all of them *30-second Elements* Eric Scerri, 2014 The chemical elements are the building blocks of life but could you discuss the periodic table over the dinner table Which elements put the blue into blue ray and the lime into limelight And do you know enough about antimony arsenic and aluminium to illuminate the bar with your elemental knowledge 30 Second Elements presents you with the foundations of chemical knowledge distilling the 50 most significant chemical elements into half a minute individual entries using nothing more than two pages 300 words and one picture Divided into seven chapters it includes the atomic details of the other 68 elements and the relationships of all 118 as well as biographies of the chemists who transformed scientific knowledge and unlocked the mysteries of life itself Illustrated with explosive graphics here is the quickest way to know your arsenic from your europium A Tale of Seven Elements Eric Scerri, 2013-07-18 In A Tale of Seven Elements Eric Scerri presents the fascinating history of those seven elements discovered to be mysteriously missing from the periodic table in 1913 **The Elements** Philip Ball, 2021-09-27 The classical elements The antique metals Alchemical elements The new metals Chemistry golden age Electrical discoveries The radiant age The nuclear age Chemistry of the Elements Norman Neill Greenwood, 1993 **What Is A Chemical Element?** Eric Scerri, Elena Ghibaudi, 2020-05-11 The concept of a chemical element is foundational within the field of chemistry but there is wide disagreement over its definition Even the International Union for Pure and Applied Chemistry IUPAC claims two distinct definitions a species of atoms versus one which identifies chemical elements with the simple substances bearing their names The double definition of elements proposed by the International Union for Pure and Applied Chemistry contrasts an abstract meaning and an operational one Nevertheless the philosophical aspects of this notion are not fully captured by the IUPAC definitions despite the fact that they were crucial for the construction of the Periodic Table Although rich scientific literature on the element and the periodic table exists as well as a recent growth in the philosophy of chemistry scholars are still searching for a definitive answer to this important question What is an element Eric Scerri and Elena Ghibaudi have teamed up to assemble a group of scholars to provide readers an overview of the current state of the debate on chemical elements from epistemological historical and educational perspectives What Is A Chemical Element fills a gap for the benefit of the whole chemistry community experimental researchers philosophers chemistry educators and anyone looking to learn more about the elements of the periodic table **Discovery of the Elements** Mary Elvira

Weeks,1956      *Chemistry of the Elements* ,2005      **The Discovery of the Periodic Table of the Chemical Elements**  
Torsten Schmiermund,2022 150 years ago in 1869 D I Mendeleev and L Meyer independently published their ideas on the arrangement of the chemical elements in a periodic system The United Nations and UNESCO therefore declared 2019 the International Year of the Periodic Table The question arises what is so special about this simple table Join the author on a short journey to the history of the periodic table Learn about its predecessors and look at how the periodic table of elements has evolved over the years Discover the periodic properties of the elements Learn what makes the periodic table so interesting and timeless and see what other ideas there are and have been for representing it The content Atomic theories atomic structure and atomic masses Early attempts to systematize the elements Mendeleev s periodic table and periodicities of the elements Alternative periodic systems The target groups Students and lecturers of chemistry especially history of chemistry Teachers and students as well as those interested in the history of chemistry The author Torsten Schmiermund has been working as a chemical technician in the chemical industry for many years This book is a translation of an original German edition The translation was done with the help of artificial intelligence machine translation by the service DeepL com A subsequent human revision was done primarily in terms of content so that the book will read stylistically differently from a conventional translation

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