

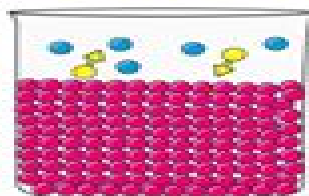
COLLIGATIVE PROPERTIES OF SOLUTIONS

A colligative property of a solution is a property that is only affected by the number of solute particles. There are four colligative properties of solutions, viz :



Vapor Pressure Drop (ΔP)

When there is a lid on the container, the gas phase molecules are trapped, they are a vapor. The vapor creates a pressure



If a solute is added to a solvent, the solute particles will prevent the solvent particles from turning into vapor. This causes a decrease in vapor pressure (ΔP).

Lots of dissolved substances $\rightarrow \Delta P$ big $\rightarrow$ low vapor pressure (P).

The amount of vapor pressure drop can be determined by the following formula

a. Non-electrolyte solution

- $$\Delta P = P^* - P$$
$$\Delta P = X_t \times P^*$$
$$\Delta P = \frac{n_t}{n_t + n_p} \times P^*$$
- $$P = X_p \times P^*$$
$$P = \frac{n_p}{n_t + n_p} \times P^*$$

b. Electrolyte Solution

The electrolyte solution in the solvent can ionize. This results in the electrolyte solution having a more significant number of particles than the non-electrolyte solution at the same concentration. The degree of ionization is a direct measure of its ionizing state. The magnitude of the degree of ionization is expressed as :

Colligative Properties Of Solutions Study Guide Answers

TEAS 6 Exambusters



Colligative Properties Of Solutions Study Guide Answers:

Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual Leo J. Malone, Theodore O. Dolter, 2012-01-03 The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment. New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections, assessment exercises at the end of each section, and relevant chapter problems at the end of each chapter. A new Math Check allows quick access to the needed basic skill. The first chapter now includes brief introductions to several fundamental chemical concepts, and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem. Every concept in the text is clearly illustrated with one or more step-by-step examples. Making it Real essays have been updated to present timely and engaging real-world applications, emphasizing the relevance of the material they are learning. This edition continues the end-of-chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter.

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