

Extension:

Osmotic pressure

Get the Gizmo ready:

- Select the STUDY SOLUTE tab.
- Set the Air temp. to 25 °C.



Introduction: **Osmosis** is the flow of water across a semipermeable membrane from a region of lower solute concentration to a region of higher solute concentration. The membrane allows water, but not solute, to go through. Like boiling and freezing, osmosis is a colligative property.

Question: How do different solutes affect osmosis?

1. **Measure:** Adding pressure to a solution can prevent water from entering the solution via osmosis. The **osmotic pressure** of the solution is equal to the pressure required to stop osmosis. A higher osmotic pressure indicates a more highly saturated solution.

Select **Sucrose** and click **Record**. What is the osmotic pressure of sucrose?

24.09

2. **Predict:** Based on the osmotic pressure of sucrose, predict what the osmotic pressures will be for the other solutes:

Sodium chloride:

28.02

Calcium chloride:

32.02

Potassium chloride:

28.02

3. **Gather data:** Use the Gizmo to measure the osmotic pressure of each solute. (Note: The **Air temp.** must be set to 25 °C to collect data on osmotic pressure.) Write the results below.

Sodium chloride:

48.18

Calcium chloride:

72.08

Potassium chloride:

48.18

4. **Analyze:** How close were your predictions?

higher than I anticipated

5. **Apply:** Eating a diet rich in sodium can increase the risk of high blood pressure. Why would excessive amounts of sodium in the blood lead to higher blood pressure?

Since there would be an excess of salt in your system, more water would want to cross the membrane into the bloodstream which could cause greater pressure.

6. **Summarize:** What do all the colligative properties you have studied have in common?

They all have an impact on pressure, freezing point, and boiling point. The quantity of particles dissolved in the solution has an impact on them.

Colligative Properties Virtual Lab Answers Key

Yicheng Fang



Colligative Properties Virtual Lab Answers Key:

Unveiling the Magic of Words: A Report on "**Colligative Properties Virtual Lab Answers Key**"

In a world defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their ability to kindle emotions, provoke contemplation, and ignite transformative change is really awe-inspiring. Enter the realm of "**Colligative Properties Virtual Lab Answers Key**," a mesmerizing literary masterpiece penned by a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve to the book is central themes, examine its distinctive writing style, and assess its profound impact on the souls of its readers.

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