

What does it mean? The last decimal place of the rock (thousandths place) should be added, but the final answer should be rounded and expressed only to the hundredths place, since the other measurement (the bucket) was only accurate to the hundredths place.

P1.7 Recognize the principle. Here we need to determine the appropriate number of significant figures when applying addition as discussed in Section 1.3.

Sketch the problem. See Figure P1.7.

Identify the relationships. When adding, the number of significant figures in the answer is determined by the value accurate to the least decimal places.

Solve. The total distance from A to C will be the distance from A to B, plus the distance from B to C.

$$\begin{array}{r} 3.45 \text{ km} \\ + 5.4 \text{ km} \quad \leftarrow \text{least accurate number (tenths place)} \\ \hline 8.9 \text{ km} \end{array}$$

What does it mean? The distances should be added including the hundredths place, but must be rounded to the tenths place for final expression because one of the values is only accurate to the tenths place.

P1.8 Recognize the principle. Keep track of significant figures in addition and subtraction. See Section 1.3.

Sketch the problem. See Figure P1.7.

Identify the relationships. The numbers should be subtracted as usual, and rounded to the smallest decimal place that is significant in all values.

Solve. The total distance from A to D will be the distance from A to B, minus the distance from B to D.

$$\begin{array}{r} 3.45 \text{ km} \\ - 3.15 \text{ km} \quad (\text{both numbers known to same accuracy}) \\ \hline 0.30 \text{ km} \end{array}$$

What does it mean? Since both measured distances are accurate to the hundredths place, we can express our answer with that accuracy as well.

P1.9 Recognize the principle. Here we need to determine the appropriate number of significant figures when applying division as discussed in Section 1.3.

Sketch the problem. No sketch is needed for this problem.

Identify the relationships. The density is defined as the mass of the object divided by its volume. Numbers should be divided as usual, then rounded to the smallest number of significant figures among all the values used.

Solve. Density is equal to $\frac{m}{V}$, so simply insert the values.

$$\begin{aligned} \text{density} &= \frac{m}{V} \\ \text{density} &= \frac{2.3 \text{ kg} \leftarrow 2 \text{ significant figures}}{0.005 \text{ m}^3 \leftarrow \text{only 1 significant figure}} \\ \text{density} &= 4.6 \times 10^3 \text{ kg/m}^3 \approx \boxed{5 \times 10^3 \text{ kg/m}^3} \end{aligned}$$

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