

CHEMISTRY FINAL EXAM REVIEW GUIDE 2012 Scott

DISCLAIMER: THIS IS A SAMPLE OF QUESTIONS/PROBLEMS FROM THIS YEAR. PLEASE BE SURE TO REVIEW YOUR NOTES, LABS, COJ'S, OBJECTIVE SHEETS, ETC. DO NOT RELY ONLY ON THIS REVIEW GUIDE.

Multiple Choice

Identify the choice that best completes the statement or answers the question.

- _____ 1) Which of the following observations is quantitative?
a. The liquid turns blue litmus paper red.
b. The liquid boils at 100°C.
c. The liquid tastes bitter.
d. The liquid is cloudy.
- _____ 2) The symbols for units of length in order from largest to smallest are
a. m, cm, mm, km. c. km, mm, cm, m.
b. mm, m, cm, km. d. km, m, cm, mm.
- _____ 3) A volume of 1 milliliter is equivalent to
a. 1 cubic centimeter. c. 1 liter.
b. 1 gram. d. 10^{-3} cubic decimeters.
- _____ 4) Which pair of quantities determines the density of a material?
a. mass and weight c. volume and concentration
b. volume and weight d. volume and mass
- _____ 5) A sample of gold has a mass of 96.5 g and a volume of 5.00 cm³. The density of gold is
a. 0.0518 g/cm³. c. 101.5 g/cm³.
b. 19.3 g/cm³. d. 483 g/cm³.
- _____ 6) The number of grams equal to 0.5 kg is
a. 0.0005. c. 500.
b. 0.005. d. 5000.
- _____ 7) If 1 inch equals 2.54 cm, how many centimeters equal 1 yard?
a. 0.0706 cm c. 30.5 cm
b. 14.2 cm d. 91.4 cm
- _____ 8) A chemist obtained the values 5.2246 g, 5.2353 g, and 5.2501 g for the mass of a sample. Without knowing the true mass of the sample, it can be said that these values have
a. good precision.
b. good accuracy.
c. poor precision.
d. poor accuracy.
- _____ 9) The measurement 0.035550 g rounded off to two significant figures would be
a. 0.03 g. c. 0.036 g.
b. 0.35 g. d. 3.5×10^2 g.
- _____ 10) When 64.4 is divided by 2.00, the correct number of significant figures in the result is

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H Kauffman



Chemistry Final Exam Study Guide Spring 2012:

Teach Students How to Learn Sandra Yancy McGuire, 2023-07-03 Co published with NISOD Miriam a freshman Calculus student at Louisiana State University made 37 5% on her first exam but 83% and 93% on the next two Matt a first year General Chemistry student at the University of Utah scored 65% and 55% on his first two exams and 95% on his third These are representative of thousands of students who decisively improved their grades by acting on the advice described in this book What is preventing your students from performing according to expectations Sandra McGuire offers a simple but profound answer If you teach students how to learn and give them simple straightforward strategies to use they can significantly increase their learning and performance For over a decade Sandra McGuire has been acclaimed for her presentations and workshops on metacognition and student learning because the tools and strategies she shares have enabled faculty to facilitate dramatic improvements in student learning and success This book encapsulates the model and ideas she has developed in the past fifteen years ideas that are being adopted by an increasing number of faculty with considerable effect The methods she proposes do not require restructuring courses or an inordinate amount of time to teach They can often be accomplished in a single session transforming students from memorizers and regurgitators to students who begin to think critically and take responsibility for their own learning Sandra McGuire takes the reader sequentially through the ideas and strategies that students need to understand and implement First she demonstrates how introducing students to metacognition and Bloom s Taxonomy reveals to them the importance of understanding how they learn and provides the lens through which they can view learning activities and measure their intellectual growth Next she presents a specific study system that can quickly empower students to maximize their learning Then she addresses the importance of dealing with emotion attitudes and motivation by suggesting ways to change students mindsets about ability and by providing a range of strategies to boost motivation and learning finally she offers guidance to faculty on partnering with campus learning centers She pays particular attention to academically unprepared students noting that the strategies she offers for this particular population are equally beneficial for all students While stressing that there are many ways to teach effectively and that readers can be flexible in picking and choosing among the strategies she presents Sandra McGuire offers the reader a step by step process for delivering the key messages of the book to students in as little as 50 minutes Free online supplements provide three slide sets and a sample video lecture This book is written primarily for faculty but will be equally useful for TAs tutors and learning center professionals For readers with no background in education or cognitive psychology the book avoids jargon and esoteric theory [Medical School Admission Requirements \(Msar\) 2011-2012](#) Aamc, Association of American Medical Colleges, 2010-03 *Applied Chemistry and Chemical Engineering, Volume 4* A. K. Haghi, Lionello Pogliani, Eduardo A. Castro, Devrim Balköse, Omari V. Mukbaniani, Chin Hua Chia, 2017-12-22 *Applied Chemistry and Chemical Engineering Volume 4* Experimental Techniques and Methodical Developments provides a detailed yet easy to follow

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