

MULTIPLE CHOICE QUESTIONS

1. When potassium dichromate crystals are heated with hydrochloric acid, we get
 (A) Oxygen (B) Hydrogen (C) Sulphur dioxide
 (D) Hydrogen sulphide (E) Chlorine Ans : E
2. Which of the following elements has lowest melting point?
 (A) Copper (B) Niobium (C) Molybdenum
 (D) Tungsten (E) Mercury Ans : E
3. In acidic medium if potassium permanganate is used as an oxidising agent, we get
 (A) MnO_4^{2-} (B) Mn^{2+} (C) MnO_2
 (D) Mn_2O_3 (E) None of these Ans : B
4. Among the following outermost configurations of transition metals, which shows the highest oxidation state?
 (A) $3d^3 4s^2$ (B) $3d^5 4s^1$ (C) $3d^6 4s^2$
 (D) $3d^6 4s^2$ (E) $3d^7 4s^2$ Ans : C
5. Highest (+7) oxidation state is shown by
 (A) Co (B) Cr (C) V
 (D) Mn (E) Ni Ans : D
6. Potassium permanganate on heating gives
 (A) $\text{K}_2\text{MnO}_4, \text{MnO}_2, \text{O}_2$ (B) $\text{MnO}_2, \text{O}_2, \text{KOH}$ (C) $\text{K}_2\text{MnO}_4, \text{O}_2$

Chemistry Multiple Choice Questions And Answers

Arshad Iqbal



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