

Acids & Bases Study Guide

6. Water is considered neutral. Describe what this means in terms of $[H^+]$ & $[OH^-]$ concentrations.

8. Complete the following:

$$pH = -\log[H^+]$$

$$K_w = [H^+][OH^-]$$

7. Fill in the blank:

9. What is the $[OH^-]$ of a solution with a pH of 10?

10. What is the $[H^+]$ of a solution with a pH of 4?

11. What is the pH of a solution with a $[H^+]$ of 1×10^{-5} M?

12. What is the pH of a solution with a $[OH^-]$ of 1×10^{-4} M?

13. What is the pH of a solution with a $[H^+]$ of 1×10^{-12} M?

14. What is the pH of a solution with a $[OH^-]$ of 1×10^{-10} M?

15. What is the pH of a solution with a $[H^+]$ of 1×10^{-11} M?

16. What is the pH of a solution with a $[OH^-]$ of 1×10^{-13} M?

Name: _____ Period: _____ Date: _____

Acids & Bases Study Guide

1. Fill in the blanks or complete and contrast the properties of Acids & Bases.

Acids	Bases
• Corrosive _____ • Protons _____ • Tastes _____ • Turn litmus paper _____ • High $[H^+]$ _____	• Corrosive _____ • Protons _____ • Tastes _____ • Turn litmus paper _____ • High $[OH^-]$ _____

2. Fill in the chart for the following acids & bases.

Acid or Base?	Strong or Weak?	Formula	Name
			hydrochloric acid
			barium hydroxide
		$HClO_4$	
		$NaOH$	
		H_2SO_4	hydrochloric acid

3. Predict the products for the following double replacement reactions involving acids & bases. Identify which component is the Acid, Base, Conjugate Acid (CA), and Conjugate Base (CB) in the formula.



4. Label the pH scale from 0 (Most Acidic) to 14 (Most Basic). Identify the pH of the following solutions.



Chem Study Guide Acids And Bases Answers

**T. W. Graham Solomons, Craig B.
Fryhle, Scott A. Snyder**



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Dolter, 2012-01-03 The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections assessment exercises at the end each section and relevant chapter problems at the end of each chapter A new Math Check allows quick access to the needed basic skill The first chapter now includes brief introductions to several fundamental chemical concepts and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem Every concept in the text is clearly illustrated with one or more step by step examples Making it Real essays have been updated to present timely and engaging real world applications emphasizing the relevance of the material they are learning This edition continues the end of chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter **Organic Chemistry, Student Study**

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Snyder, 2016-04-11 This is the Student Study Guide Solutions Manual to accompany Organic Chemistry 12th Edition The 12th

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